

Nail Art (6 total hours)

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Lesson 1:Anatomy of the Nail (2 hours)

Outline

- Parts of the Nail
- Nail Disorders
- Nail Conditions
- Nail Diseases
- Product Safety

Learning objectives:

After completing this lesson you will be able to:

- list and define the parts of the nail
- identify nail disorders, conditions, and infections
- explain the effects of disorders and conditions
- define white spots, splinter hemorrhages, ingrown nails and fungal infections
- describe germ science
- define pathogenic and non-pathogenic bacteria
- define viral infections
- identify parasites: protozoa helminths and ectoparasites
- explain blood-borne parasites
- describe how nail conditions can cause changes in the nail's appearance
- define contagious disorders
- explain how to protect yourself in a chemical environment
- describe how to reduce inhalation, skin exposure, and accidental swallowing of chemical products
- identify chemical products and symptoms of exposure

Introduction

The purpose of this study is to review nail anatomy, nail conditions and product safety. It is important for the professional manicurist to be aware of all factors relating to the nail and it's surrounding tissues in order to perform nail services in a safe and healthy manor.

The Nail

The nail in it's created form has several functions of the human body. It protects and supports the surrounding tissues of the fingers and toes. Fingernails help us to have more dexterity in our interactions. If we did not have fingernails we would have trouble picking up objects or doing other daily tasks.

Another important function that the nail possesses is that it shows our body's level of health. **The appearance of the nail changes when there are negative health issues going on in our bodies.** This helps physicians to identify certain health issues in their patients.

Another function of the nail is an acquired function. It is cosmetic appearance. As a nation, Americans spend a lot of money on the cosmetic appearance of their nails. Both men and women in modern societies often put significant time and effort into maintaining the appearance and health of their nails.

Nails are created by living skin cells in the fingers and toes. Nails are primarily keratin. Keratin is a type of human protein.

Growth

The nail is created from the matrix area. There is a pattern of nail growth in that new cells are created and followed by more new cells. The nail area, as it progresses toward the free edge contain older and older cells. The cells at the free edge are the oldest cells. Toenails do not grow as fast as the nails of the fingers. Nails also grow according to seasons, for example in the winter, nail growth slows down.

Parts of the Nail:

Cuticle	the non-living skin that adheres to the nail plate
Distal Phalanges	the fingertip bones that gives shape to the nail unit and supports the dermis
Eponychium	the skin that covers the newly developing nail plate
Hyponychium	the epidermis under the free edge of the nail plate that forms the water tight seal
Keratin	the hard protein that the matrix produces
Lateral Nail Folds	the sidewall seals along each side of the nail plate
Lunula	the blueish white half-moon shape at the base of the nail, usually most pronounced on the thumb
Matrix	the area under the eponychium, the hidden part of the nail where growth takes place
Nail Bed	the skin beneath the nail plate
Nail Folds	the folds of skin that frame and support the nail on three sides
Nail Plate	the visible part of the nail on fingers and toes made of up of hard keratin
Onychodermal Band	the area between the bed epithelium and the solehorn that bunches together to form a band of delicate tissue - it borders the white free edge under the nail plate.
Proximal Nail fold	the fold nearest the point where the nail attaches.
Solehorn Cuticle	a thin layer of epithelium that attaches to the underside of the nail plate at the free edge and naturally sloughs away as the nail grows

Common Nail Disorders

Medical science places nail disorders into the skin conditions category. Nail disorders make up about 10 % of all skin conditions. Because people are very active with their hands and feet, the nail is highly susceptible to injury and in some cases infection. The nail will show signs of problems if injured or diseased. It can change shapes as it grows, it can turn different colors and it can also develop dents or ridges. Change in appearance may indicate a minor health problem. Some changes could indicate a more serious health issue.

White Spots

When the base of the nail is injured, white spots can develop within the nail. The white spots are not harmful and will eventually grow out. This is a very common condition and it reoccurs frequently once they develop. They are most often in the shape of circles or semi-circles.

Splinter Hemorrhages

When blood vessels burst inside the nail bed it can cause what is called splinter hemorrhages. The bleeding or hemorrhaging area runs vertical within the nail and they take on the look of splinters, thus called splinter hemorrhages. This condition is most often caused by trauma to the nail however certain medications and diseases can also cause this condition.

Ingrown Nails

Ingrown nails form at the corner of the nail. The nail most often begins to grow downward and curve into the skin due to repeated pushing or pressure. It happens most often to the large toenail. This condition can be brought about by improperly trimming the nails as well. Tight shoes or improper stance can bring about this condition. Ingrown nails can be painful and sometimes become infected.

Fungal Infections

Fungal infections can occur to nails of the feet or hands. Fungal infections often cause the end of the nail plate to separate from the nail bed. The nail can become discolored. The top of the nail and the skin at the base of the nail can also become infected and become very painful. **Toenails are more susceptible to fungal infections than fingernails** because of the moist, warm environment. Fungi thrive where moisture and warmth are present. **Fifty percent of all nail disorders are fungal infections.**

Other types of infections can occur in the nail. Some examples of common infections are candida and yeast. Persons whose job keeps their hands wet are more susceptible to this type of infection on their fingernails. Other bodily health issues also raise the level of susceptibility for these infections to occur.

Discoloration and change in nail shape can also occur with this type of infection. Fungal infection of the nail, or onychomycosis, is often ignored because the infection can be present for years without causing any pain. Onychomycosis can also be accompanied by a secondary bacterial or yeast infection in or around the nail plate. Because it is difficult to avoid contact with microscopic organisms like fungi, the toenails are especially vulnerable around damp areas where people are likely to be walking barefoot, such as swimming pools, locker rooms, and showers.

Prevention

- Keep feet and toes clean and inspect them often.
- Keep the feet dry.
- After washing the feet and toes with soap and water, dry thoroughly.
- Wear shower shoes when in public areas such as water parks, pools, spas .
- Change your hosiery more than once daily.
- Trim toenails properly .
- Wear shoes that fit well and are made of materials that allow air exchange.
- Disinfect instruments used to cut nails.
- Don't apply polish to nails suspected of infection

Bacterial Infections

Bacterial infections are recognizable due to the redness and swelling that they cause.

The bacteria begins to grow and can cause great pain in the nail skin folds. Bacteria sets up quickly if there is an injury in the nail area or to the skin surrounding the nail.

Bacterial infections can cause the nail to turn green.

Germ Science

Bacteriology is the scientific study of bacteria. Bacteria is responsible for some of the most deadly diseases known to man. Diseases such as tuberculosis, typhus, plague, diphtheria, typhoid fever, cholera, dysentery, and pneumonia are all caused by bacterial infections. **Bacteria can live on the hands and nails and can be spread through contact with others or inanimate objects.**

An essential part of preventing the spread of infection in the salon is proper hygiene and sanitation. This includes hand-washing and properly disinfecting implements and surfaces.

Non Pathogenic Bacteria

Non pathogenic means that the bacteria is incapable of causing disease.

Non pathogenic bacteria are harmless.

Pathogenic Bacteria

Pathogenic means that the bacteria does cause disease. Bacteria is capable of causing diseases in humans, in animals, and also in plants.

Viruses

Most infections are caused by either bacteria or viruses.

Influenza is an example of a viral infection. It infects the lungs and is accompanied by fever, cough, and severe muscle aches.

There are three types of influenza

Influenza A - Responsible for regular outbreaks. Influenza A viruses also infect domestic animals like pigs, horses, chickens, ducks and birds.

Influenza B - Often causes sporadic outbreaks of illness, especially in residential communities like nursing homes.

Influenza C - Common but seldom causes disease symptoms.

Viral Infections

Warts are the results of a viral infection. Warts can be found in any portion of the nail area. The nail plate could change shape or be destroyed due to interference of a growing wart. Warts can be painful.

Parasites

A parasite is an organism that lives on or in a host.

There are three main classes of parasites that can cause disease in humans:

Protozoa

Helminths and

Ectoparasites

Protozoa are microscopic, one-celled organisms that can be free-living or parasitic in nature. They are able to multiply in humans and can cause serious infections to develop from just a single organism.

Helminths are large, multi-cellular organisms that are generally visible to the naked eye in their adult stages. Like protozoa, helminths can be either free-living or parasitic in nature.

Ectoparasites

Although the term ectoparasites can include a broad range of blood-sucking arthropods such as mosquitoes, this term is generally used to refer to ticks, fleas, lice, and mites that attach or burrow into the skin.

Blood-borne Parasites

Some parasites can be blood-borne. This means two things:

1. The parasite sometimes can be found in the blood stream of infected people; and
2. The parasite might be spread to other people through exposure to an infected person's blood, for example, by blood transfusion or by sharing needles or syringes contaminated with blood.

Appearance of the Nail

The most common health conditions that change the appearance of the nail:

<u>Condition</u>	<u>Nail Appearance</u>
Anemia	Pale nail beds
Diabetes	Yellowish nails, with a slight blush at the base
Heart Conditions	Nail bed is red
Kidney Disease	Half of nail is pink, half is white
Liver Disease	White Nails
Lung Diseases	Yellowing and thickening of the nail, slowed growth rate

The appearance of nails can show the initial sign of the presence or onset of systemic diseases. For example, the pitting of nails and increased nail thickness can be manifestations of psoriasis.

Concavity

Nails that are growing in a concave shape (rounded inward) can be because of an iron deficiency in the person.

Contagious Disorders

Athlete's Foot	Ringworm of the foot.
Tinea	Ringworm caused by fungus, a vegetable parasite, that includes symptoms of scaling of the skin.
Tinea Unguium	Ringworm of the nails.

Nail Diseases/Disorders

Agnail	The hangnail.
Beau's Lines	Ridges, corrugations and furrows of the nail.
Eggshell Nail	Very thin nail.
Hematoma Nail	A bruised nail.

Leuconychia	The disorder where there are white spots under the nail plate.
Onychatrophia	The atrophy or wasting away of the nail.
Onychauxis	The overgrowth of the nail plate.
Onychia	Inflammation of the nail.
Onychocryptosis	An ingrown nail.
Onychocyanosis	When the nail is blue due to poor circulation.
Onychogryposis	When the nail is extremely curved like a claw.
Onychophagy	Nail biting.
Onychophosis	The accumulation of horny layers of epidermis under the nail.
Onychosis	Refers to any nail disease.
Paronychia	The inflammation of tissue due to bacteria around the nail.
Tinea Corporis	Ringworm of the hand.
Tinea Pedia	Ringworm of the foot.
Tinea Unguium	Ringworm of the nail, also known as onychomycosis.

Protecting the Manicurist and Pedicurist

Nail Technicians provide a very important and popular service to clients. There are tens of thousands of licensed professionals across America who perform manicures and pedicures daily in hundreds of salons.

Services include nail care, hand care, foot care, manicures, pedicures, nail polishing and the application of artificial nails.

Technical services require the use of many formulations of key products. These products include dynamic active ingredients and can be hazardous to the health of the professional and to their client if not handled properly and professionally.

The minimization to exposure to these ingredients, especially inhalation, must be the highest level of concern for the salon professional.

Overall, product manufacturers have formulated products to help minimize exposure and have tried to lessen negative health effects for the user and client. When products are used in smaller amounts it can minimize exposure.

Even though manufacturers consider exposure when developing their products, it is the responsibility of the product user to do what is necessary to protect themselves and their clients as much as possible during their use.

Injury can happen if improperly handled. Injury can occur to the eyes, nose and lungs. There can also be allergic reactions and sensitivities to certain products or ingredients that only until after this reaction happens that you are aware that you have an allergy.

The Environmental Protection Agency protects the salon professional and their clients by setting regulations.

Decreasing the Inhalation of Salon Products

An exhaust ventilation system must be in place near all working tables of the manicurist. This is in addition to any air conditioning system or heating system that is a standard modern part of the salon.

We must capture and expel vapors, nail filings and chemical powders and dust away from the clients and manicurists.

Tips for Decreasing Inhalation:

- The use of ceiling exhaust systems
- The use of table exhaust systems that are vented to the outside of the building
- The use of air conditioning systems
- The regular replacement of air filters in your air conditioning and heating systems
- The use of electronic air cleaners
- Not using excessive amounts of products when performing services
- Keeping products in small containers at the work table
- Keeping the work table clear of obstructions so your table exhaust system works optimally
- Keeping containers closed tightly containers when not in use
- Not using bulk product containers while performing a service
- Keeping lids on trash cans, emptying trash cans frequently and replacing liners often
- Wearing a dust mask while transferring products from one container to another

Decreasing Skin Exposure to Salon Products

- Wash hands before and after touching a patron or performing a service
- Wash hands before eating or going to the bathroom
- Wash hands after handling products or product containers
- Wear disposable nitrile gloves
- Replace gloves as soon as they are damaged or compromised in any way
- Keep containers tightly closed
- Do not wear clothing that is too loose and could cause accidents with sleeves or other areas getting caught on equipment
- Not using excessive amounts of products when performing services
- Keeping products in small containers at the work table
- Wearing eye protection and gloves while transferring products from one container to another

Prevention of Accidental Swallowing of Salon Products

- Wash hands every time before eating or drinking.
- Never eat or drink in service areas.
- Do not have food or drinks in service areas.
- Do not have salon products in food areas.
- Keep clear and separate areas for eating and working.

A Clean Environment

- Sanitize the salon.
- Dispose of gloves after each customer.
- Use clean unused towels for each customer.
- Clean implements before disinfecting them.
- Disinfect implements after each customer.
- Never give services to unhealthy nails, broken skin or any other signs of unsafe health.
- Before giving the service wash your patrons hands and/or feet.
- Do not use implements brought by customers
- Do not use razor-type shavers to remove calluses.
- Follow your State Cosmetology Board's rules for proper cleaning and disinfection techniques

Safe Practices for the Manicurist

- Label all containers
- Only dispose of waste in accordance to manufacturers' directions or with State regulations
- Do not use outdated MSDS
- Keep copies of MSDS in a binder for each product in your salon that requires it
- Keep copies of state board rules and regulations
- Be trained concerning OSHA rules.
- No smoking by you or patrons inside the salon building.

Important Facts for the Manicurist

- ◆ Wearing gloves keeps manicurists from damaging their nails and skin
- ◆ Wearing gloves keeps the skin from absorbing ingredients that could cause harm or irritation or allergic reactions to their hands.
- ◆ Solvents can be absorbed very quickly by the hands which can be extremely harmful.
- ◆ When solvents damage the skin it causes the damaged area to be more susceptible to further absorption and damage by other harmful products.
- ◆ Only nitrile gloves provide sufficient protection due to the fact that many product ingredients can penetrate them.
- ◆ Look at the MSDS for instructions on what kind of gloves to wear or contact the distributor of the product.
- ◆ Wear dust masks to keep from inhaling dust particles
- ◆ Wear dust masks when shaping artificial nails or filing natural nails

- ◆ Dust masks are approved by the National Institute for Occupational Safety and Health
- ◆ There are specific instructions on how to use themselves
- ◆ MSDS will advise on which type of mask is best

Product Ingredients

Symptoms of Overexposure

Acetone nail polish remover and fingernail glue remover	Headache, dizziness, irritation to skin, eyes, and throat
Benzoyl peroxide powder additive for artificial nails	Irritation to eyes, mouth, throat, nose, and lungs, and skin rash
Butyl acetate nail polish	Irritation to skin, eyes, mouth, nose and throat; skin rash, headache, drowsiness and confusion
Butyl methacrylate artificial nails	Irritation to skin, eyes, skin, mouth, nose, and throat, skin rash, and shortness of breath
Camphor nail polish	Irritation to skin, eyes, mouth, nose and throat, nausea, vomiting, diarrhea, headache, dizziness, and in extreme cases of overexposure, uncontrollable muscle contractions
Dibutyl phthalate nail polish, nail hardener	Irritation to eyes, stomach, and upper respiratory system
Ethyl acetate nail polish, fingernail glue	Irritation to skin, eyes, mouth, nose and throat, skin rash and confusion
Ethyl cyanoacrylate fingernail glue	Irritation to skin, eyes, mucous membranes and skin sensitization
Ethyl methacrylate artificial nails	Irritation to skin, eyes, respiratory track and skin sensitization
Formalin nail hardener	Irritation to skin, eyes, nose, throat and respiratory system, and wheezing.

Lesson 1 Summary

We have now reviewed basic information regarding the Anatomy of the Nail. We have discussed parts of the nail, disorders, conditions, diseases and product safety. Let's now turn our attention to the next topic: Creative Artistry.

Lesson 2: Creative Artistry (2 hours)

Outline

- Nail Art Services
- Color Schemes
- Polish
- French Manicure
- Warm and Cool Colors
- The Color Wheel
- Gems
- Foil Methods
- Striping Tape
- Gold Leafing
- Free-hand Painting
- Types of Brushes
- Parts of the Brush
- Air Brushing

Learning objectives:

After completing this lesson you will be able to:

- describe color as it applies to nail art
- identify methods of creating color schemes
- identify nail free-hand painting
- describe the french manicure and the reverse french manicure
- identify the colors within the warm and cool color pallets
- describe the color wheel and how it applies to nail art
- describe types of nail gems and their application
- identify types of foiling material and methods of application
- define the types and applications of striping tape
- describe types of gold leafing materials and the methods of application
- identify free-hand painting
- identify types of nail art brushes
- identify parts of a brush
- define air brushing

Introduction

The purpose of this study is to review principles and practices of Nail Art to include Nail Services, Nail Art Implements and Nail Art Application Procedures.

As you read the following material, think of how you would apply the previous information to these services. Ask yourself how each of the following services can be performed safely and in a healthy manner as you review their procedures. Consider how you would evaluate the health of the patron's nails, feet and hands in order to perform the following services.



Color schemes

Color schemes are logical combinations of colors of the color wheel. Color schemes are used to create style and appeal. Colors that create a pleasing appearance commonly appear together in color schemes.

A basic color scheme uses two colors that look appealing together. More advanced color schemes involve several colors in combination, usually based around a single color.

For example: fingernails with such colors as **red**, **yellow**, **orange** and light **blue** arranged together on a black background.

Color schemes can also contain different shades of a single color; for example, a color scheme that mixes **different shades of green**, ranging from very light to very dark.

Polish

Traditional colors for nail polish are **red**, **pink** and **brown**, but nail polish is manufactured in many colors.

French Manicure

The French manicure is traditionally patterned after the color of natural nails, using a clear, beige or soft pink polish on most of the nail with a white finish at the tips.

Reverse French manicures are worn as well but not as often as the traditional French manicure.

Warm and Cool Colors

Warm colors always contain **golden** undertones.

Cool colors always contain **blue** undertones.

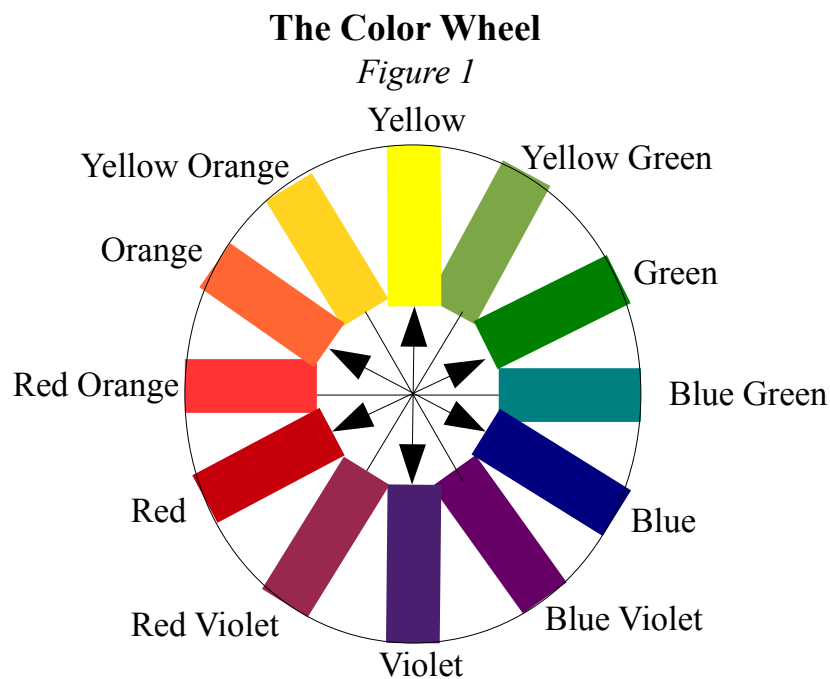
Nail technicians often use color palettes that stay within the same category. Either all of the colors in the palette will be cool, or all of the colors in the palette will be warm.

Most professionals agree that staying within one category creates a more esthetically appealing look. Using colors from opposing categories tends to create a more contrasting look with less cohesiveness in appearance.

Pallets of color can be created by applying the theory that certain color combinations work well together and appear harmonious.

The word pallet has several definitions. **Pallet is a term for color combinations used by an artist** and it is a term for the actual device that holds the paint while the artist is creating.

The color wheel is an irreplaceable tool in developing color pallets for the most beautiful nail artistry.



One method of developing a triad color scheme is to choose three colors equidistant around the hue circle.

The color wheel is one of the best tools in color pallet determination. Choosing colors of equal depth, in other words, equal levels of pigment, can be a favorable choice when a balanced appearance is desired.

Using contrasting color choices for your pallet can create an appearance of multiple dimensions within the artwork.

Pre-planned color pallets can be presented to the customer. The choices can range from vibrant colors to muted soft colors.

Choosing the overall background color should be the first step in developing a color pallet. Then choosing a secondary dominate color is usually the next step, then third, fourth, and so on.

Occasionally, the customer may want to match a dominate color in her wardrobe for her background nail color.

On the other hand, neutral and natural nail colorings are popular choices making it completely unnecessary to match any wardrobe colors. In fact, the number one choice in nail color is neutral, natural and beige or light pink tones.

Gems

Tiny gems are manufactured specifically for nail art. They come in various shapes and sizes. They are specifically designed with one side cut flat for attaching to the nail surface.

Apply the gem to the nail with top coat or nail art sealer. **Place sealer on the flat part of the gem so it will adhere to the nail.**

You may also apply the gem with a wooden pusher. First dampen the end of the pusher with the sealer.

Pick up the gem by touching the dampened stick to the colorful side of the gem and place it in the prepared spot on the nail. You can use tweezers as well to do this task.

Press the gem onto the nail and apply a generous amount of sealer to coat it.

Gems can be removed with acetone and can be reused if the gem is in good condition.

Foiling

Foiling is easy and lends opportunity for endless creativity. **Foil comes in rolls and pre-cut pieces in many colors and designs.** Some of the most popular are silver, gold and snake skin. Pre-cut pieces are easy to handle and convenient.

Before applying foil, polish the nail and allow to completely dry. Select the polish color as part of the color scheme.

Apply foil adhesive to the dry nail. Apply the adhesive thinly and evenly. The adhesive is slightly tinted when wet and becomes clear and tacky when dry. This tackiness is similar to double sided tape and is developed for easy adhesion to foil.

Foil comes attached to a cellophane base, so when the foil is touched to the tacky nail, it pulls off of the cellophane and adheres to the nail. Wait until the adhesive becomes clear, but do not wait too long or the adhesive will become overly dry and lose it's stickiness.

The timing of the tacky phase is imperative to the success of the service.

There is a shiny side and a matte side to the foil. **The shiny side is always to be facing up and the matte side is to be against the adhesive.**

Foil Methods

Method one is a type of free hand designing. The nail technician pats the cellophane, foil side down, onto the nail. The cellophane is clear after the foil is removed from it, so the technician watches for coverage areas.

The nail technician is careful not to touch the clear portions of the cellophane to the nail. If the nail is touched with the clear portion of the cellophane instead of the foil area, it is possible that the cellophane would stick to the adhesive which could in turn remove the adhesive along with the nail polish.

Method two involves complete coverage. The foil is gently laid on the surface of the nail and a wooden pusher is used to lightly burnish or lightly rub over the cellophane backing, applying the foil to the nail. This is the method used for pre-manufactured foil patterns.

All nail art applications require that you use a final coat of nail art sealer. To apply you must “float the bead”. In other words, drop a “bead” or drop of sealant onto the nail and pull the liquid gently to cover the entire nail. Do not brush using pressure as you would polish, but touch the “top” of the bead of dropped sealant and disperse it gently in that manner.

Striping Tape

Striping tape can be applied as the first design application after the nail polish is dry or it can be applied as the last design application when all other design features are in place and are dry.

Striping tape is manufactured in rolls and pages with many colors and sizes. The most used colors are gold, silver, and black. **Striping tape has a tacky backing.**

Gold Leafing

Leafing material is very thin and fragile. It has a foil-like consistency and is manufactured both in sheets and in loose pieces. Loose gold leafing is also referred to as nuggets and the gold leaf sheets are referred to as nugget sheets.

Leafing sheets are very delicate and must be handled with tweezers or the tissue paper that comes between the sheets in the packaging, otherwise they can crumble or blow away. Usually sheets of leafing come in quantities of 10 to 100 sheets per package. Keep the package closed or the slightest air movement can blow away the product.

The Application Method

Apply nail adhesive to the dry polished nail and allow to become tacky.

Apply the leafing in small amounts using tweezers or a dampened wooden pusher.

When applying a sheet, use the same method as foiling.

When leafing is applied in pieces it appears more like nuggets, and when applied in sheets you get a thin even coverage.

Complete the application with nail art sealer.

Free Hand Painting

Freehand painting is also referred to as flat nail art.

It is when you use nail polish and brushes to create art design on the nail surface.

Brushes

To the artist, the proper choice and quality of brushes and nail color play a crucial part in creating precision designs. Brushes come in many shapes and sizes and many varieties of bristle types. There are firm bristles and soft bristles, slender shapes and thicker shapes. **A complete knowledge of color theory is required to create the most desirable works of art.**

Parts of a Brush

The tip of the brush is the end of the bristles farthest away from the handle. It is also called the edge of the brush.

The mid section of the bristles is called **the belly**. This is where the brush holds the most paint.

The ferrule is the metal band around the brush that helps to hold the bristles in place.

The heel of the brush is the area where the bristles meet the ferrule.

Knowing the parts of the brush is the beginning of understanding why certain brush choices are made to achieve the desired result.

Round brushes have pointed tapered tips and a large belly.

Round brushes vary in their tips making some better for detail work than others.

Softness of the bristles or hairs of the brush determine the amount of control in detail work as well.

Liner brushes are thin and pointed.

They are great for lettering or other fine work requiring detail.

They are best, for example, drawing thin lines and outlining other art applications.

Flat brushes are flexible with long bristles and a square tip.

Another name for a flat brush is shader brush.

It has a flat tip or chiseled edge.

It is used for multi-purposes.

Using the flat of the brush creates smooth strokes.

Each side of the brush can be loaded with different colors to create fade effects. This is called double loading.

The best way to understand it's versatility is to practice designs on art paper to see the limitless creations from this one brush.

Bright brushes are short and flat with very firm bristles.

They are great for creating texture.

Fan brushes are flat brushes in which the bristles are spread out like a fan.

It's a great special effects tool and also can be used for blending. I

t's great for special effects.

When used dry with just a dab of paint on the bristle ends, it can be floated across the top of a design and look similar to air brushing.

Spotter brushes are also called a detailer.

It is short and round with a small belly and a very fine point tip.

It is excellent for the most intricate of detail work.

Striper brushes are very long flat brushes with few bristles.

It's great for creating designs that need long lines or stripes.

The stripette brush also known as the short steeper brush is simply a shorter version of the steeper brush and it creates the same effects.

The marbleizer brush also known as a stylus comes with wooden handles, a rounded tip and a metal extension. They have rounded ball tips that come in different sizes. They are used for dotting small circles of color such as polka dots or bubbles. A marbleized effect can be created by using different colored paints in a swirl motion similar to the look of marble.

Air Brushing

Airbrushing has become a phenomenally popular salon service.

The technique of color fading can easily be achieved with the airbrush.

Airbrushing is used for the French manicure to achieve the perfect white edge.

A stencil is used to define the shape for the desired end results of the white nail tip.

By using airbrushes and stencils you can easily apply images onto the nail. Pre-cut stencils are readily available at all professional supply stores. **Stencils are made of either plastic, paper or fabric.** You can create your own stencils.

Airbrush Equipment

Airbrushes are usually classified by three characteristics.

The first characteristic is the action performed by the user to trigger the nail color flow.

The second is the mechanism for feeding the nail color into the airbrush.

The third is the point at which the nail color and air mix.

Trigger

The simplest airbrushes work with a single action mechanism where the depression of a single trigger results in nail color and air flowing into the airbrush body and the atomized paint being expelled onto the target surface. Dual action or double action airbrushes separate the function for air and nail color flow so that the user can control the volume of airflow and the concentration of nail color flow through two independent mechanisms.

Technique

Airbrush technique is the freehand manipulation of the airbrush, nail color, air pressure and distance from the surface being sprayed in order to produce a certain predictable result on a consistent basis with or without shields or stencils. Airbrush technique will differ with the type of airbrush being used. Read the manufacturers instructions for operating your airbrush. Practice on paper to become accomplished in technique. See what works and what does not. There are specific practices to master in developing beautiful work. Practice proper aim. Practice on nail tips mounted on wooden sticks. It's a great way to become accustomed to the exact thing of what you are trying to perfect, the nail. Practice painting straight lines, dots and with stencils.

Review The Following Terminology

Nail Art

belly	midsection of the brush bristles; the area that holds the most nail color
color wheel	color guide that illustrates and identifies the primary, secondary, tertiary and complementary colors
double loading	placing two different colors of paint on either side of the brush
floating the bead	technique used to seal nail art where a bead of sealer is dropped onto the nail surface, and the brush floats across the surface and completely covers it with sealer
foil adhesive	special adhesive just for foiling that is generally tinted white or pink and appears cloudy when it is wet
freehand painting	using no stencils or pre-formed patterns, creating art on nails also known as flat nail art
French manicure	airbrushing technique that creates a natural looking nail with a smooth white tip at the free edge
gem	tiny jewel added to a nail design, attached to the nail with adhesive
heel	the part of the brush where the bristles of the brush meet the ferrule
gold leafing	also known as nuggets or nugget sheets, foil like material used to create nail art
marbleizer	also known as stylus, tool with a rounded ball tip and wooden handle use for making dots or small circles
stencil	precut design made of plastic, paper or fabric used to create nail art
striping tape	tacky-backed tape available in various colors; applied over nail polish or other nail art to create bolder designs

Lesson 2 Summary

In this lesson we have discussed facts regarding Creative Artistry, We have reviewed practices of Nail Art Services and have discussed the importance of Color Schemes. We understand methods in polishing, French manicures and have reviewed warm and cool colors. We know the importance of the Color Wheel and have reviewed facts about gems, foil methods, striping tape and Gold Leafing. We have reviewed Free-hand Painting and the types of brushes used. And lastly we have discussed facts about Air Brushing. It is important to continue learning new methods of nail artistry and to also rely on existing methods.

Lesson 3: Health, Safety, and Welfare (2 hours)

Part 1: Sanitation and Sterilization

Outline

- Standard cleaning and disinfecting precautions
- How to distinguish between disinfectants and antiseptics
- How to sanitize hands and disinfect tools used in the practice of cosmetology
- Bacterial, viral, and fungal control
- Bloodborne pathogens control
- Parasite infection and infestation control

Learning objectives

After completing this lesson you will be able to

- identify cleaning precautions
- describe cleaning routines
- describe the importance of reading labels
- define the term clean
- describe the properties of ideal disinfectants
- identify sterilization
- describe wet disinfection
- define the purpose of infection control
- list safety and sanitation requirements
- describe cleaning and disinfecting procedures
- identify salon environment requirements
- identify CDC infection control guidelines
- describe chlorine based disinfectants
- describe phenol based disinfectants
- describe QAC based disinfectants
- list types of sterilization techniques
- identify lice and infestation

Introduction

In this chapter we will identify standard cleaning and disinfecting precautions, disinfectants and antiseptics, hand sanitation, tool disinfection, bacterial, viral, and fungal control, bloodborne pathogens control, and parasite infection and infestation control.

Standard Cleaning and Disinfecting Precautions

Protect Yourself

We must take precautions and heed the warnings of all labels regarding the handling, use and storage of cleaning chemicals. We will go more into an in-depth study of specific safety issues when we study OSHA regulations later in this course. Just bear in mind that we must observe all safety rules and regulations at all times when preparing, mixing, and applying chemicals. Follow all label instructions without alteration. Only use chemicals designed for the specific material you are cleaning, otherwise damage may occur.

Protect Your Clients

There are Federal and State Guidelines that the personal service worker must adhere to in order to protect each and every client as well as the salon worker.

How should cleaners and disinfectants be used?

Read the label first. Each cleaner and disinfectant has instructions on the label that tell you important facts:

- How to apply the product to a surface.
- How long you need to leave it on the surface to be effective (contact time).
- If the surface needs to be cleaned first and rinsed after using.
- If the disinfectant is safe for the surface.
- Whether the product requires dilution with water before use.
- Precautions you should take when applying the product, such as wearing gloves or aprons or making sure you have **good ventilation** during application.

Facility Cleaning & Disinfection

Cleaning and disinfection should be performed on surfaces that are likely to contact your patron.

- *Cleaning surfaces with detergent-based cleaners or Environmental Protection Agency (EPA)-registered disinfectants is effective at removing germs, viruses and fungi from the environment.*
- It is important to read the instruction labels on all cleaners to make sure they are used safely and appropriately.
- Environmental cleaners and disinfectants should not be used to treat the skin
- The EPA provides a list of EPA-registered products



Illustration 1: Cleaning surfaces

Surfaces to Clean

Focus on surfaces that touch people's bare skin each day and any surfaces that could come into contact with people. Clean large surfaces such as floors and walls when they become visibly soiled or on a regular schedule. There is no evidence that spraying or fogging rooms or surfaces with disinfectants will prevent infections more effectively than the targeted approach of cleaning frequently touched surfaces and any surfaces that have been exposed to infections.

Shared Equipment

Shared equipment that comes into direct skin contact should be cleaned after each use and allowed to dry. Salon equipment, and protective gear, should also be cleaned according to the equipment manufacturers' instructions to make sure the cleaner will not harm the item.

Cleaning Keyboards and other Difficult Surfaces

Many items such as computer keyboards or handheld electronic devices may be difficult to clean or disinfect or they could be damaged if they became wet. If these items are touched by many people during the course of the day, a cleanable cover/skin could be used on the item to allow for cleaning while protecting the item. Always check to see if the manufacturer has instructions for cleaning.

Laundry

Routine laundry procedures, detergents, and laundry additives will all help to make clothes, towels, and linens safe to wear or touch. If items have been contaminated by infectious material, these may be laundered separately, but this is not absolutely necessary.

Proper Water Temperature for Laundry

Read and follow the clothing and soap or detergent label instructions. Water temperatures for laundry depend on the type of fiber or fabric. In general, wash and dry in the warmest temperatures recommended on the fabric label. Also, some laundry detergents are made to clean best at certain temperatures. Not following instructions could damage the item or decrease the effectiveness of the detergent.

Using Bleach for Laundry

Use of bleach as a disinfectant in laundering is optional, and not all fabrics are suitable for bleach. Read the label instructions.

Routines

- Facilities should always be kept clean.
- Review cleaning procedures and schedules with the staff.
 - Cleaning procedures should focus on commonly touched surfaces and surfaces that come into direct contact with people's bare skin each day.
 - Cleaning with detergent-based cleaners or Environmental Protection Agency (EPA)-registered detergents/disinfectants will remove bacteria from surfaces.
 - Cleaners and disinfectants, including household chlorine bleach, can be irritating and exposure to these chemicals has been associated with health problems such as asthma and skin and eye irritation.
 - ❑ Take appropriate precautions described on the product's label instructions to reduce exposure. Wearing personal protective equipment such as gloves and eye protection may be indicated.
 - Follow the instruction labels on all cleaners and disinfectants, including household chlorine bleach, to make sure they are used safely and correctly.
 - ❑ Some key questions that should be answered by reading the label include:
 - ❑ How should the cleaner or disinfectant be applied?
 - ❑ Do you need to clean the surface first before using the disinfectant (e.g., precleaned surfaces)?
 - ❑ Is it safe for the surface? Some cleaners and disinfectants, including household chlorine bleach, might damage some surfaces (e.g., metals, some plastics).
 - ❑ How long do you need to leave it on the surface to be effective (i.e., contact time)?
 - ❑ Do you need to rinse the surface with water after using the cleaner or disinfectant?
 - If you are using household chlorine bleach, check the label to see if the product has specific instructions for disinfection.
 - Environmental cleaners and disinfectants should not be put onto skin or wounds and should never be used to treat infections.
- Repair or dispose of equipment and furniture with damaged surfaces that do not allow surfaces to be adequately cleaned.

How to distinguish between disinfectants and antiseptics, How to sanitize hands and disinfect tools used in the practice of cosmetology, Bacterial, viral, and fungal control, Bloodborne pathogens control, and Parasite infection and infestation control

First, let's review disinfectants

What's the difference between cleaners, sanitizers, and disinfectants?

- **Cleaners or detergents** are products that are used to remove soil, dirt, dust, organic matter, and germs (like **bacteria, viruses, and fungi**). Cleaners or detergents work by washing the surface to lift dirt and germs off surfaces so they can be rinsed away with water. The same thing happens when you wash your hands with soap and water or when you wash dishes. Rinsing is an important part of the cleaning process. Use these products for routine cleaning of surfaces.
- **Sanitizers** are used to reduce **germs** from surfaces but not totally get rid of them. Sanitizers reduce the germs from surfaces to levels that are considered safe.
- **Disinfectants** are chemical products that **destroy or inactivate germs** and prevent them from growing. Disinfectants have no effect on dirt, soil, or dust. Disinfectants are regulated by the U.S. Environmental Protection Agency (EPA).

CDC – Infection Control Guidelines

The Centers for Disease Control and Prevention (CDC) have developed Guidelines for Disinfection and Sterilization.

They provide the following information:

Cleaning is the removal of foreign material (e.g., soil, and organic material) from objects and is normally accomplished using water with detergents or enzymatic products. Thorough cleaning is required before high-level disinfection and sterilization because inorganic and organic materials that remain on the surfaces of salon implements and equipment interfere with the effectiveness of disinfection and sterilization.

Also, if soiled materials dry, the removal process becomes more difficult and the disinfection or sterilization process less effective or ineffective.

With manual cleaning, the two essential components are friction and soap/detergents and water.

Friction (e.g., rubbing/scrubbing the soiled area with a brush) is an old and dependable method. Soap/detergents and water is used to remove soil and debris.



Illustration 2: Wash with soap and water

Disinfection

Many chemical disinfectants are used alone or in combinations. These include chlorine and chlorine compounds, phenolics, and quaternary ammonium compounds. Commercial formulations based on these chemicals are considered unique products and must be registered with EPA or cleared by FDA. In most instances, a given product is designed for a specific purpose and is to be used in a certain manner. Therefore, users should read labels carefully to ensure the correct product is selected for the intended use and applied efficiently.

Disinfectants are not interchangeable, and incorrect concentrations and inappropriate disinfectants can result in excessive costs. Because occupational diseases have been associated with use of several disinfectants such as chlorine, precautions (e.g., gloves and proper ventilation) should be used to minimize exposure.

Asthma and reactive airway disease can occur in sensitized persons exposed to any airborne chemical, including germicides.

Clinically important asthma can occur at levels below ceiling levels regulated by OSHA (Occupational Safety and Health Administration) or recommended by NIOSH (the National Institute for Occupational Safety and Health).

The following is information that can be used to select an appropriate disinfectant for any item and use it in the most efficient way.

Properties of an ideal disinfectant

- Broad spectrum: should have a wide antimicrobial spectrum
- Fast acting: should produce a rapid kill
- Not affected by environmental factors: should be active in the presence of organic matter (e.g., blood, sputum, feces) and compatible with soaps, detergents, and other chemicals encountered in use
- Nontoxic: should not be harmful to the user or patron
- Surface compatibility: should not corrode instruments and metallic surfaces and should not cause the deterioration of cloth, rubber, plastics, and other materials
- Residual effect on treated surfaces: should leave an antimicrobial film on the treated surface
- Easy to use with clear label directions
- Odorless: should have a pleasant odor or no odor to facilitate its routine use
- Economical: should not be prohibitively high in cost
- Solubility: should be soluble in water
- Stability: should be stable in concentrate and use-dilution
- Cleaner: should have good cleaning properties
- Environmentally friendly: should not damage the environment on disposal

Barbicide

Barbicide is a disinfectant solution used by barbers and cosmetologists for sterilizing grooming tools such as combs and hair-cutting shears. Manufactured by King Research, it was invented in 1947 by Maurice King and marketed heavily around the United States by his brother James.

Barbicide is a United States Environmental Protection Agency approved combination germicide, pseudomonacide, fungicide, and a viricide effective against the HIV-1 virus (AIDS virus), Hepatitis B, and Hepatitis C. Its active ingredient is Alkyl dimethyl benzyl ammonium chloride (5.12% by volume); sodium nitrite and blue dye are also present. Contact can cause irritation to the skin and eyes, and consumption of 50 mL can cause shock and may lead to death if not treated quickly.

Barbicide is sold as a concentrate diluted for use in a 1:32 ratio with water, with each stylist having a container for treating their own tools. At one time, several US states legally required barber shops to use Barbicide; according to the maker two still did in 1997. A jar of Barbicide sits on display in the Smithsonian Institution's National Museum of American History.



*Illustration 3:
Barbicide jar*

Chlorine and Chlorine Compounds



Illustration 4: Bleach

Hypochlorites, the most widely used of the chlorine disinfectants, are available as liquid (e.g., sodium hypochlorite) or solid (e.g., calcium hypochlorite).

The most prevalent chlorine products in the United States are aqueous solutions of 5.25%–6.15% **sodium hypochlorite, usually called household bleach**. They have a broad spectrum of antimicrobial activity, do not leave toxic residues, are unaffected by water hardness, are inexpensive and fast acting, remove dried or fixed organisms and biofilms from surfaces, and have a low incidence of serious toxicity. Sodium hypochlorite at the concentration

used in household bleach (5.25-6.15%) can produce ocular irritation or oropharyngeal, esophageal, and gastric burns. Other disadvantages of hypochlorites include corrosiveness to metals in high concentrations (>500 ppm), inactivation by organic matter, discoloring or “bleaching” of fabrics, and the **release of toxic chlorine gas when mixed with ammonia or other household cleaning agents**.

EPA has determined the currently registered uses of hypochlorites will not result in unreasonable adverse effects to the environment.

Phenols



Illustration 5:
Lysol concentrate

Phenol has occupied a prominent place in the field of disinfection since its initial use as a germicide by Lister in his pioneering work on antiseptic surgery. In the past 30 years, however, work has concentrated on the numerous phenol derivatives or phenolics and their antimicrobial properties. Phenol derivatives originate when a functional group (e.g., alkyl, phenyl, benzyl, halogen) replaces one of the hydrogen atoms on the aromatic ring. Two phenol derivatives commonly found as constituents of disinfectants are *ortho*-phenylphenol and *ortho*-benzyl-*para*-chlorophenol. The antimicrobial properties of these compounds and many other phenol derivatives are much improved over those of the parent chemical. Phenolics are absorbed by porous materials, and the residual disinfectant can irritate the skin.

Examples of phenol disinfectant cleaners are Pine-Sol and Lysol.

Microbicidal Activity of Phenols

Published reports on the antimicrobial efficacy of commonly used phenolics showed they were bactericidal, fungicidal, virucidal, and tuberculocidal at their recommended use-dilution.

Quaternary Ammonium Compounds

Quaternary Ammonium Compounds (QACs) are a type of chemical that is used to kill bacteria, viruses, and mold. QACs are widely used as disinfectants. The quaternaries are good cleaning agents, but high water hardness and materials such as cotton and gauze pads can make them less microbicidal because of insoluble precipitates or cotton and gauze pads absorb the active ingredients, respectively .



Illustration 6:
Quaternary disinfectant

Examples of QAC products are Lysol Spray and Clorox Disinfectant Spray.

Microbicidal Activity of QACs

Results from manufacturers' data sheets and from published scientific literature indicate that the quaternaries sold as hospital grade disinfectants are generally fungicidal, bactericidal, and virucidal against most viruses.

Salon Computer Keyboards

Quaternary ammonium compounds (as well as 70% isopropyl alcohol, phenolic, and a chlorine-containing wipes effectively (>95%) remove and/or inactivate contaminants from computer keyboards with a 5-second application time. No functional damage or cosmetic changes occurred to the computer keyboards after 300 applications of the disinfectants.

EPA and FDA

In the United States, chemical germicides formulated as sanitizers, disinfectants, or sterilants are regulated in interstate commerce by the Antimicrobials Division, Office of Pesticides Program, EPA, under the authority of the Federal Insecticide, Fungicide, and Rodenticide Act (FIFRA) of 1947, as amended.

Under FIFRA, any substance or mixture of substances intended to prevent, destroy, repel, or mitigate any pest (including microorganisms but excluding those in or on living humans or animals) must be registered before sale or distribution.

To obtain a registration, a manufacturer must submit specific data about the safety and effectiveness of each product. For example, EPA requires manufacturers of sanitizers, disinfectants, or chemical sterilants to test formulations by using accepted methods for microbiocidal activity, stability, and toxicity to animals and humans.

The manufacturers submit these data to EPA along with proposed labeling. If EPA concludes the product can be used without causing "unreasonable adverse effects," then the product and its labeling are registered, and the manufacturer can sell and distribute the product in the United States.

Sterilization

Sterilization destroys all microorganisms on the surface of an article or in a fluid to prevent disease transmission associated with the use of that item. The use of inadequately sterilized items represents a high risk of transmitting pathogens.

The concept of what constitutes "sterile" is measured as a probability of sterility for each item to be sterilized.

This probability is commonly referred to as the sterility assurance level (SAL) of the product and is defined as the probability of a single viable microorganism occurring on a product after sterilization. SAL is normally expressed as 10^{-6} .

For example, if the probability of a spore surviving were one in one million, the SAL would be 10^{-6} . In short, a SAL is an estimate of lethality of the entire sterilization process and is a conservative calculation.

Liquid Chemicals

Several FDA-cleared liquid chemical sterilants include indications for sterilization of implements. The indicated contact times range from 3 hours to 12 hours. These solutions are commonly used as high-level disinfectants when a shorter processing time is required.

Ultraviolet Radiation (UV)

UV radiation has been used in the disinfection of drinking water, air, and contact lenses. Bacteria and viruses are more easily killed by UV light than are bacterial spores.

The wavelength of UV radiation ranges from 328 nm to 210 nm (3280 Å to 2100 Å). Its maximum bactericidal effect occurs at 240–280 nm. Mercury vapor lamps emit more than 90% of their radiation at 253.7 nm, which is near the maximum microbicidal activity. Inactivation of microorganisms results from destruction of nucleic acid through induction of thymine dimers.



Illustration 7: UV Sterilizer

Hand-hygiene

Washing hands at key times with soap and water is one of the most important steps you can take to get rid of germs and avoid spreading germs to those around you.



Illustration 8: Washing hands

How can washing your hands keep you healthy?

Germs can get into the body through our eyes, nose, and mouth and make us sick.

Handwashing with soap removes germs from hands and helps prevent sickness. Studies have shown that handwashing can prevent 1 in 3 diarrhea-related sicknesses and 1 in 5 respiratory infections, such as a cold or the flu.

Handwashing helps prevent infections for these reasons:

People often touch their eyes, nose, and mouth without realizing it, introducing germs into their bodies. Germs from unwashed hands may get into foods and drinks when people prepare or consume them. Germs can grow in some types of foods or drinks and make people sick. Germs from unwashed hands can be transferred to other objects, such as door knobs, tables, or toys, and then transferred to another person's hands.

What is the correct way to wash your hands?

1. Wet your hands with clean running water (warm or cold) and apply soap.
2. Lather your hands by rubbing them together with the soap.
3. Scrub all surfaces of your hands, including the palms, backs, fingers, between your fingers, and under your nails. Keep scrubbing for at least 20 seconds. Need a timer? Hum the “Happy Birthday” song twice.
4. Rinse your hands under clean, running water.
5. Dry your hands using a clean towel or air dry them.

When should you wash your hands?

- Before, during, and after preparing food
- Before eating food
- Before and after caring for someone who is sick
- Before and after treating a cut or wound
- After using the bathroom, changing diapers, or cleaning up a child who has used the bathroom
- After blowing your nose, coughing, or sneezing
- After touching an animal, animal food or treats, animal cages, or animal feces (poop)
- After touching garbage
- If your hands are visibly dirty or greasy

What type of soap should you use?

You can use bar soap or liquid soap to wash your hands. Many public places provide liquid soap because it's easier and cleaner to share with others. Studies have not found any added health benefit from using soaps containing antibacterial ingredients when compared with plain soap. Both are equally effective in getting rid of germs. If soap and water are not available, use an alcohol-based hand sanitizer that contains at least 60% alcohol.

Hand Sanitizers

Washing hands with soap and water is the best way to reduce the number of germs on them in most situations. If soap and water are not available, use an alcohol-based hand sanitizer that contains at least 60% alcohol. Alcohol-based hand sanitizers can quickly reduce the number of germs on hands in some situations, but sanitizers do **not** eliminate all types of germs and might not remove harmful chemicals.

Hand sanitizers are not as effective when hands are visibly dirty or greasy.

How do you use hand sanitizers?

- Apply the product to the palm of one hand (read the label to learn the correct amount).
- Rub your hands together.
- Rub the product over all surfaces of your hands and fingers until your hands are dry.

Nail Hygiene

Appropriate hand hygiene includes diligently cleaning fingernails, which may harbor dirt and germs and can contribute to the spread of some infections. Before clipping or grooming nails, all equipment should be properly cleaned. Sterilizing equipment before use is especially important. Infections of the fingernails are often characterized by swelling of the surrounding skin, pain in the surrounding area, or thickening of the nail. In some cases, these infections may be serious and need to be treated by a physician.

To help prevent the spread of germs and nail infections:

- Scrub the underside of nails with soap and water (or a nail brush) every time you wash your hands.

- Clean any nail grooming tools before use.
- Sterilize nail grooming tools before use.
- Avoid biting or chewing nails.
- Avoid cutting cuticles, as they act as barriers to prevent infection.
- Never rip or bite a hangnail. Instead, clip it with a clean, sanitized nail trimmer.

Global Handwashing Day

Celebrate Global Handwashing Day to promote handwashing with soap throughout the world. Global Handwashing Day is a way to support a global and local culture of handwashing with soap, shine a spotlight on the act of handwashing in each country, and raise awareness about the benefits of handwashing with soap. Since 2008, Global Handwashing Day has been celebrated annually on **October 15** worldwide. The Global Public-Private Partnership for Handwashing with Soap founded Global Handwashing Day and encourages school children, teachers, and families to get involved.

People and communities around the world will celebrate Global Handwashing day in many ways, including:

- Learning how to wash hands the right way through an online video produced by the Centers for Disease Control
- Watching a Facebook Live talk on why handwashing with soap is so important.
- Sharing handwashing lessons, events, and materials for thousands of students across the state of Georgia, where CDC is based.

How does handwashing help fight antibiotic resistance?

Antibiotic resistance occurs when bacteria resist the effects of an antibiotic – that is, germs are not killed and they continue to grow. Sickesses caused by antibiotic-resistant bacteria can be harder to treat. Simply using antibiotics creates resistance, so avoiding infections in the first place reduces the amount of antibiotics that have to be used and reduces the likelihood that resistance will develop during treatment. Handwashing helps prevent many sicknesses, meaning less use of antibiotics.

Antiseptics

Dictionary.com defines antiseptics as: A substance that inhibits the proliferation of infectious microorganisms. Proliferation is the growth or production of cells by multiplication of parts.

Google explains them this way: antiseptic (anti-sep-tik) n. a chemical, such as **chlorhexidine** or **cetrimide**, that destroys or inhibits the growth of disease-causing bacteria and other microorganisms. Antiseptics are used externally to cleanse wounds and internally to treat infections of the intestine and bladder.

Encyclopedia.com explains antiseptics as:

An antiseptic is a substance that inhibits the growth and development of microorganisms. For practical purposes, antiseptics are routinely thought of as topical agents, for application to skin, mucous membranes, and inanimate objects, although a formal definition includes agents that are used internally, such as the urinary tract antiseptics.

Purpose

Antiseptics are a diverse class of drugs that are applied to skin surfaces or mucous membranes for their anti-infective effects.

This may be either bacteriocidal (kills bacteria) or bacteriostatic (stops the growth of bacteria). Their uses include cleansing of skin and wound surfaces after injury, preparation of skin surfaces prior to injections or surgical procedures, and routine disinfection of the oral cavity as part of a program of oral hygiene. Antiseptics are also used for disinfection of inanimate objects, including instruments and furniture surfaces.

Commonly used antiseptics for skin cleaning include benzalkonium chloride, chlorhexidine, hexachlorophene, alcohol, and hydrogen peroxide.

Other agents that have been used for this purpose, but have largely been supplanted by more effective or safer agents, include boric acid and volatile oils such as methyl salicylate (oil of wintergreen).

Chlorhexidine shows a high margin of safety when applied to mucous membranes, and has been used in oral rinses and preoperative total body washes.

Benzalkonium chloride and hexachlorophene are used primarily as hand scrubs or face washes. Benzalkonium may also find application as a disinfecting agent for instruments, and in low concentration as a preservative for drugs including ophthalmic solutions. Benzalkonium chloride is inactivated by organic compounds, including soap, and must not be applied to areas that have not been fully rinsed.

Hydrogen peroxide acts through the liberation of oxygen gas. Although the antibacterial activity of hydrogen peroxide is relatively weak, the liberation of oxygen bubbles produces an effervescent action, which may be useful for wound cleansing through removal of tissue debris. The activity of hydrogen peroxide may be reduced by the presence of blood and pus. The appropriate concentration of hydrogen peroxide for antiseptic use is 3%, although higher concentrations are available.

Precautions

Precautions vary with individual product and use. Hypersensitivity reactions should be considered with organic compounds such as chlorhexidine, benzalkonium and hexachlorophene. Skin dryness and irritation should be considered with all products, but particularly with those containing alcohol.

Most antiseptics have not been rated according to pregnancy category under the pregnancy risk factor system.

Hexachlorophene is schedule C during pregnancy, and should not be used on newborns due to risk of systemic absorption with potential central nervous system (CNS) effects, including convulsions.

Application of hexachlorophene to open wounds, mucous membranes, or areas of thin skin, such as the genitalia, should be avoided, since this may promote systemic absorption.

Chlorhexidine should not be instilled into the ear. There is one anecdotal report of deafness following use of chlorhexidine in a patient with a perforated eardrum. Safety in pregnancy and breastfeeding have not been reported; however there is one anecdotal report of an infant developing slowed heartbeat apparently related to maternal use of chlorhexidine.

Interactions

Antiseptics are not known to interact with any other skin products. However, they should not be used together with any other topical cream, solution, or ointment.

Parasite infection

Lice infestation

A lice infestation, or pediculosis, is caused by parasites living on human skin. Lice are tiny, wingless insects with sucking mouthparts that feed on human blood and lay eggs on body hair or in clothing. Lice bites can cause intense itching.

There are three related species of human lice:

- head lice, *Pediculus humanus capitis*
- body lice, *Pediculus humanus corporis*
- pubic lice, *Phthirus pubis*, commonly called crab lice

Pediculosis capitis is an infestation of head lice. A body lice infestation is called pediculosis corporis. Pediculosis palpebrarum or phthiriasis palpebrarum, caused by crab lice, is an infestation of the pubic hair.

Head lice live and crawl on the scalp, sucking blood every three to six hours. Their claws are adapted for clinging to hair or clothing. Adult head lice can be silvery-white to reddish-brown. They are about the size of a sesame seed. Female lice lay their eggs in sacs called nits that are about 0.04 in (1 mm) long and are glued to shafts of hair close to the scalp. During her one-month lifespan a female louse may lay more than 100 eggs. The nymphs hatch in three to 14 days and must feed on blood within one day. Nymphs are smaller and lighter in color than adults and become sexually mature after nine to 12 days.

Body lice lay their nits in clothing or bedding. Occasionally the nits are attached to body hair. Body lice nits are oval and yellow to white in color. They may not hatch for up to 30 days. Nymphs mature in about seven days.

Pubic lice have large front legs and look like tiny crabs. Females are larger than males. Nits hatch in about one week and the nymphs mature in about seven days.

Transmission

Lice are endemic in human populations, spreading through personal contact or contact with infested clothing or other personal items. They can be transmitted when unaffected clothing is stored with infested items. Among children head lice are commonly transmitted by the sharing of hats, combs, brushes, hair accessories, headphones, pillows, and stuffed **toys**. Pubic lice are sexually transmitted, although occasionally they can be transmitted through infested bedding, towels, or clothing.

Lice do not jump, hop, or fly and they do not live on pets. Head lice cannot survive without a human host for more than a few days at most. Body lice can live without human contact for up to 10 days. Pubic lice can survive for one to two weeks without human contact.

Head lice infestations are extremely common among children in schools, childcare facilities, camps, and playgrounds. They are the second most common communicable health problem in children, after the common cold, and appear to be on the increase. Some 6 to 12 million American children get head lice every year. In developing countries more than 50 percent of the general population may be infested. Although anyone can get head lice, children aged three to ten and their families are most affected.

Lice infestations are characterized by intense itching caused by an allergic reaction to a toxin in lice saliva. The itching can interfere with sleep and concentration. Repeated bites can lead to generalized skin eruptions or inflammation. Swelling or inflammation of the neck glands are common complications of head lice.

Prevention

Prevention of lice infestation depends on adequate personal hygiene and consistently not sharing combs, brushes, hair accessories, hats, towels, or bedding. Hair should be checked weekly for lice and nits. Prevention includes sanitation and sterilization of salon equipment and implements as directed using an EPA approved product.

Part 1 Summary

In this lesson we have thoroughly discussed the most important aspects of standard cleaning and disinfecting precautions, how to distinguish between disinfectants and antiseptics, and how to sanitize hands and disinfect tools used in the practice of cosmetology. We now understand how to control the spread of bacterial, viral, and fungal infections. We have also reviewed practices in bloodborne pathogens control as well as parasite infection and infestation control. Personal care service workers must consistently strive to maintain professional sanitation and sterilization methods in order to protect themselves and others in the salon environment.

In the following segment, we will discuss OSHA regulations.

Health, Safety, and Welfare

Part 2: Occupational Safety and Health Administration Regulations

Outline

- Hazard Communication Standard
- Safety Data Sheets
- Manufacturer's SDS
- Safety In The Workplace
- Employer's Best Practices

Learning objectives

After completing this lesson you will be able to

- list the 16 sections of Safety Data Sheets
- identify the purpose of the Hazard Communication Standard
- list the required sections and its contents of a Safety Data Sheet
- recognize a manufacturer's SDS
- describe the ways employers are required to provide a safe workplace
- identify aspects of filing a complaint
- explain employer responsibilities in maintaining a safe workplace
- list the action steps in pre-assessment of hazards
- describe aspects of personal protective equipment

Introduction

With the Occupational Safety and Health Act of 1970, Congress created the Occupational Safety and Health Administration (OSHA) to assure safe and healthful working conditions for working men and women by setting and enforcing standards and by providing training, outreach, education and assistance. OSHA is part of the United States Department of Labor. The administrator for OSHA is the Assistant Secretary of Labor for Occupational Safety and Health. OSHA's administrator answers to the Secretary of Labor, who is a member of the cabinet of the President of the United States.

Hazard Communication Standard

What is Hazard Classification?

Hazard classification is the process of evaluating the full range of available scientific evidence to determine if a chemical is hazardous, as well as to identify the level of severity of the hazardous effect. When complete, the evaluation identifies the hazard class(es) and associated hazard category of the chemical. The HCS defines hazard class as the nature of a physical or health hazard, e.g., flammable solid, carcinogen, and acute toxicity.

Hazard Communication Standard

OSHA's Hazard Communication Standard (HCS) is designed to protect against chemical injuries and illnesses by ensuring that employers and workers are provided with sufficient information to anticipate, recognize, evaluate, and control chemical hazards and take appropriate protective measures. This information is provided through **safety data sheets (SDSs), labels, and employee training. In order for SDSs, labels, and training to be effective, the hazard information they convey must be complete and accurate.** Thus, it is critically important to obtain comprehensive and correct information about the hazards associated with particular chemicals.

SAFETY DATA SHEETS (SDS)

Sections 1 through 8 contain general information about the chemical, identification, hazards, composition, safe handling practices, and emergency control measures (e.g., fire fighting). This information should be helpful to those that need to get the information quickly.

Sections 9 through 11 and 16 contain other technical and scientific information, such as physical and chemical properties, stability and reactivity information, toxicological information, exposure control information, and other information including the date of preparation or last revision. The SDS must also state that no applicable information was found when the preparer does not find relevant information for any required element.

Sections 12 through 15, is required to be consistent with the UN Globally Harmonized System of Classification and Labeling of Chemicals (GHS), but OSHA will not enforce the content of these sections because they concern matters handled by other agencies.

A description of all 16 sections of the SDS, along with their contents, is presented below:

Section 1: Identification

This section identifies the chemical on the SDS as well as the recommended uses. It also provides the essential contact information of the supplier. The required information consists of:

- Product identifier used on the label and any other common names or synonyms by which the substance is known.
- Name, address, phone number of the manufacturer, importer, or other responsible party, and emergency phone number.
- Recommended use of the chemical (e.g., a brief description of what it actually does, such as flame retardant) and any restrictions on use (including recommendations given by the supplier).

Section 2: Hazard(s) Identification

This section identifies the hazards of the chemical presented on the SDS and the appropriate warning information associated with those hazards. The required information consists of:

- The hazard classification of the chemical (e.g., flammable liquid, category¹).
- Signal word.
- Hazard statement(s).
- Pictograms (the pictograms or hazard symbols may be presented as graphical reproductions of the symbols in black and white or be a description of the name of the symbol (e.g., skull and crossbones, flame).
- Precautionary statement(s).
- Description of any hazards not otherwise classified.
- For a mixture that contains an ingredient(s) with unknown toxicity, a statement describing how much (percentage) of the mixture consists of ingredient(s) with unknown acute toxicity. Please note that this is a total percentage of the mixture and not tied to the individual ingredient(s).

Section 3: Composition/Information on Ingredients

This section identifies the ingredient(s) contained in the product indicated on the SDS, including impurities and stabilizing additives. This section includes information on substances, mixtures, and all chemicals where a trade secret is claimed. The required information consists of:

Substances

- Chemical name.
- Common name and synonyms.
- Chemical Abstracts Service (CAS) number and other unique identifiers.
- Impurities and stabilizing additives, which are themselves classified and which contribute to the classification of the chemical.

Mixtures

- Same information required for substances.
- The chemical name and concentration (i.e., exact percentage) of all ingredients which are classified as health hazards and are:
 - Present above their cut-off/concentration limits or
 - Present a health risk below the cut-off/concentration limits.
- The concentration (exact percentages) of each ingredient must be specified except concentration ranges may be used in the following situations:
 - A trade secret claim is made,
 - There is batch-to-batch variation, or
 - The SDS is used for a group of substantially similar mixtures.

Chemicals where a trade secret is claimed

- A statement that the specific chemical identity and/or exact percentage (concentration) of composition has been withheld as a trade secret is required.

Section 4: First-Aid Measures

This section describes the initial care that should be given by untrained responders to an individual who has been exposed to the chemical. The required information consists of:

- Necessary first-aid instructions by relevant routes of exposure (inhalation, skin and eye contact, and ingestion).
- Description of the most important symptoms or effects, and any symptoms that are acute or delayed.
- Recommendations for immediate medical care and special treatment needed, when necessary.

Section 5: Fire-Fighting Measures

This section provides recommendations for fighting a fire caused by the chemical. The required information consists of:

- Recommendations of suitable extinguishing equipment, and information about extinguishing equipment that is not appropriate for a particular situation.
- Advice on specific hazards that develop from the chemical during the fire, such as any hazardous combustion products created when the chemical burns.
- Recommendations on special protective equipment or precautions for firefighters.

Section 6: Accidental Release Measures

This section provides recommendations on the appropriate response to spills, leaks, or releases, including containment and cleanup practices to prevent or minimize exposure to people, properties, or

the environment. It may also include recommendations distinguishing between responses for large and small spills where the spill volume has a significant impact on the hazard. The required information may consist of recommendations for:

- Use of personal precautions (such as removal of ignition sources or providing sufficient ventilation) and protective equipment to prevent the contamination of skin, eyes, and clothing.
- Emergency procedures, including instructions for evacuations, consulting experts when needed, and appropriate protective clothing.
- Methods and materials used for containment (e.g., covering the drains and capping procedures).
- Cleanup procedures (e.g., appropriate techniques for neutralization, decontamination, cleaning or vacuuming; adsorbent materials; and/or equipment required for containment/clean up)

Section 7: Handling and Storage

This section provides guidance on the safe handling practices and conditions for safe storage of chemicals. The required information consists of:

- Precautions for safe handling, including recommendations for handling incompatible chemicals, minimizing the release of the chemical into the environment, and providing advice on general hygiene practices (e.g., eating, drinking, and smoking in work areas is prohibited).
- Recommendations on the conditions for safe storage, including any incompatibilities. Provide advice on specific storage requirements (e.g., ventilation requirements)

Section 8: Exposure Controls/Personal Protection

This section indicates the exposure limits, engineering controls, and personal protective measures that can be used to minimize worker exposure. The required information consists of:

- OSHA Permissible Exposure Limits (PELs), American Conference of Governmental Industrial Hygienists (ACGIH) Threshold Limit Values (TLVs), and any other exposure limit used or recommended by the chemical manufacturer, importer, or employer preparing the safety data sheet, where available.
- Appropriate engineering controls (e.g., use local exhaust ventilation, or use only in an enclosed system).
- Recommendations for personal protective measures to prevent illness or injury from exposure to chemicals, such as personal protective equipment (PPE) (e.g., appropriate types of eye, face, skin or respiratory protection needed based on hazards and potential exposure).
- Any special requirements for PPE, protective clothing or respirators

Section 9: Physical and Chemical Properties

This section identifies physical and chemical properties associated with the substance or mixture. The minimum required information consists of:

- Appearance (physical state, color, etc.);
- Upper/lower flammability or explosive limits;
- Odor;
- Vapor pressure;
- Odor threshold;
- Vapor density;

- pH;
- Relative density;
- Melting point/freezing point;
- Solubility(ies);
- Initial boiling point and boiling range;
- Flash point;
- Evaporation rate;
- Flammability (solid, gas);
- Partition coefficient: n-octanol/water;
- Auto-ignition temperature;
- Decomposition temperature; and
- Viscosity.

The SDS may not contain every item on the above list because information may not be relevant or is not available. When this occurs, a notation to that effect must be made for that chemical property. Manufacturers may also add other relevant properties, such as the dust deflagration index (Kst) for combustible dust, used to evaluate a dust's explosive potential

Section 10: Stability and Reactivity

This section describes the reactivity hazards of the chemical and the chemical stability information. This section is broken into three parts: reactivity, chemical stability, and other. The required information consists of:

Reactivity

- Description of the specific test data for the chemical(s). This data can be for a class or family of the chemical if such data adequately represent the anticipated hazard of the chemical(s), where available.

Chemical stability

- Indication of whether the chemical is stable or unstable under normal ambient temperature and conditions while in storage and being handled.
- Description of any stabilizers that may be needed to maintain chemical stability.
- Indication of any safety issues that may arise should the product change in physical appearance.

Other

- Indication of the possibility of hazardous reactions, including a statement whether the chemical will react or polymerize, which could release excess pressure or heat, or create other hazardous conditions. Also, a description of the conditions under which hazardous reactions may occur.
- List of all conditions that should be avoided (e.g., static discharge, shock, vibrations, or environmental conditions that may lead to hazardous conditions).
- List of all classes of incompatible materials (e.g., classes of chemicals or specific substances) with which the chemical could react to produce a hazardous situation.
- List of any known or anticipated hazardous decomposition products that could be produced because of use, storage, or heating. (Hazardous combustion products should also be included in Section 5 (Fire-Fighting Measures) of the SDS.)

Section 11: Toxicological Information

This section identifies toxicological and health effects information or indicates that such data are not available. The required information consists of:

- Information on the likely routes of exposure (inhalation, ingestion, skin and eye contact). The SDS should indicate if the information is unknown.
- Description of the delayed, immediate, or chronic effects from short- and long-term exposure.
- The numerical measures of toxicity (e.g., acute toxicity estimates such as the LD50 (median lethal dose)) - the estimated amount [of a substance] expected to kill 50% of test animals in a single dose.
- Description of the symptoms. This description includes the symptoms associated with exposure to the chemical including symptoms from the lowest to the most severe exposure.
- Indication of whether the chemical is listed in the National Toxicology Program (NTP) Report on Carcinogens (latest edition) or has been found to be a potential carcinogen in the International Agency for Research on Cancer (IARC) Monographs (latest editions) or found to be a potential carcinogen by OSHA

Section 12: Ecological Information (non-mandatory)

This section provides information to evaluate the environmental impact of the chemical(s) if it were released to the environment. The information may include:

- Data from toxicity tests performed on aquatic and/or terrestrial organisms, where available (e.g., acute or chronic aquatic toxicity data for fish, algae, crustaceans, and other plants; toxicity data on birds, bees, plants).
- Whether there is a potential for the chemical to persist and degrade in the environment either through biodegradation or other processes, such as oxidation or hydrolysis.
- Results of tests of bioaccumulation potential, making reference to the octanol-water partition coefficient (K_{ow}) and the bioconcentration factor (BCF), where available.
- The potential for a substance to move from the soil to the groundwater (indicate results from adsorption studies or leaching studies).
- Other adverse effects (e.g., environmental fate, ozone layer depletion potential, photochemical ozone creation potential, endocrine disrupting potential, and/or global warming potential).

Section 13: Disposal Considerations (non-mandatory)

This section provides guidance on proper disposal practices, recycling or reclamation of the chemical(s) or its container, and safe handling practices.

To minimize exposure, this section should also refer the reader to Section 8 (Exposure Controls/Personal Protection) of the SDS. The information may include:

- Description of appropriate disposal containers to use.
- Recommendations of appropriate disposal methods to employ.
- Description of the physical and chemical properties that may affect disposal activities.
- Language discouraging sewage disposal.
- Any special precautions for landfills or incineration activities

Section 14: Transport Information (non-mandatory)

This section provides guidance on classification information for shipping and transporting of hazardous chemical(s) by road, air, rail, or sea. The information may include:

- UN number (i.e., four-figure identification number of the substance)¹.
- UN proper shipping name¹.
- Transport hazard class(es)¹.
- Packing group number, if applicable, based on the degree of hazard².
- Environmental hazards (e.g., identify if it is a marine pollutant according to the International Maritime Dangerous Goods Code (IMDG Code)).
- Guidance on transport in bulk (according to Annex II of MARPOL 73/78³ and the International Code for the Construction and Equipment of Ships Carrying Dangerous Chemicals in Bulk (International Bulk Chemical Code (IBC Code))).
- Any special precautions which an employee should be aware of or needs to comply with, in connection with transport or conveyance either within or outside their premises (indicate when information is not available).

Section 15: Regulatory Information (non-mandatory)

This section identifies the safety, health, and environmental regulations specific for the product that is not indicated anywhere else on the SDS. The information may include:

- Any national and/or regional regulatory information of the chemical or mixtures (including any OSHA, Department of Transportation, Environmental Protection Agency, or Consumer Product Safety Commission regulations)

Section 16: Other Information

This section indicates when the SDS was prepared or when the last known revision was made. The SDS may also state where the changes have been made to the previous version. You may wish to contact the supplier for an explanation of the changes. Other useful information also may be included here.

Who Must Conduct Hazard Classifications?

Only chemical manufacturers and importers are required to perform hazard classifications on the chemicals they produce or import. Under the HCS, an employer that manufactures, processes, formulates, blends, mixes, repackages, or otherwise changes the composition of a hazardous chemical is considered a "chemical manufacturer."

The following 16 page document is the official
manufacturer's Safety Data Sheet
for Lysol Brand Disinfectant Concentrate

SAFETY DATA SHEET

Lysol Brand Disinfectant Concentrate



HEALTH • HYGIENE • HOME

1. Product and company identification

Product name : Lysol Brand Disinfectant Concentrate

Distributed by : Reckitt Benckiser LLC.
Morris Corporate Center IV
399 Interpace Parkway (P.O. Box 225)
Parsippany, New Jersey 07054-0225
+1 973 404 2600

Emergency telephone number (Medical) : 1-800-338-6167

Emergency telephone number (Transport) : 1-800-424-9300 (U.S. & Canada) CHEMTREC
Outside U.S. and Canada (North America), call Chemtrec:703-527-3887

Website: <http://www.rbnainfo.com>

Product use : Disinfectant.

This SDS is designed for workplace employees, emergency personnel and for other conditions and situations where there is greater potential for large-scale or prolonged exposure, in accordance with the requirements of USDOL Occupational Safety and Health Administration.

This SDS is not applicable for consumer use of our products. For consumer use, all precautionary and first aid language is provided on the product label in accordance with the applicable government regulations, and shown in Section 15 of this SDS.

SDS # : 353773PSDS v3.0

Formulation #: : 269-005 (353773 v10.0)

EPA ID No. : 777-94

UPC Code / Sizes : 19200-02201-10; 19200-77500-10 (12 fl.oz. PET Amber Pour Bottle with CRC cap)

2. Hazards identification

Classification of the substance or mixture : FLAMMABLE LIQUIDS - Category 4
SKIN CORROSION/IRRITATION - Category 1C
SERIOUS EYE DAMAGE/ EYE IRRITATION - Category 1

GHS label elements

Hazard pictograms :



Signal word : Danger

Hazard statements : Combustible liquid.
Causes severe skin burns and eye damage.

Precautionary statements

Code # : EE353773
COURSE: Nail Art (6 total hours)
(353773PSDS)

SDS # : 353773PSDS v3.0

Date of issue : 01/04/2015.

1/16

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2. Hazards identification

- General** : Read label before use. Keep out of reach of children. If medical advice is needed, have product container or label at hand.
- Prevention** : Wear protective gloves. Wear eye or face protection. Wear protective clothing. Keep away from flames and hot surfaces. - No smoking. Wash hands thoroughly after handling.
- Response** : IF INHALED: Remove victim to fresh air and keep at rest in a position comfortable for breathing. Immediately call a POISON CENTER or physician. IF SWALLOWED: Immediately call a POISON CENTER or physician. Rinse mouth. Do NOT induce vomiting. IF ON SKIN (or hair): Take off immediately all contaminated clothing. Rinse skin with water or shower. Wash contaminated clothing before reuse. Immediately call a POISON CENTER or physician. IF IN EYES: Rinse cautiously with water for several minutes. Remove contact lenses, if present and easy to do. Continue rinsing. Immediately call a POISON CENTER or physician.
- Storage** : Store locked up. Store in a well-ventilated place. Keep cool.
- Disposal** : Dispose of contents and container in accordance with all local, regional, national and international regulations.
- Supplemental label elements** : None known.
- Hazards not otherwise classified** : None known.

3. Composition/information on ingredients

Substance/mixture : Mixture

Ingredient name	%	CAS number
clorofene	5 - 10	120-32-1
potassium hydroxide	2.5 - 5	1310-58-3
Ethyl alcohol	1 - 2.5	64-17-5
Isopropyl alcohol	1 - 2.5	67-63-0

Any concentration shown as a range is to protect confidentiality or is due to batch variation.

There are no additional ingredients present which, within the current knowledge of the supplier and in the concentrations applicable, are classified as hazardous to health or the environment and hence require reporting in this section.

4. First aid measures

Description of necessary first aid measures

- Eye contact** : Get medical attention immediately. Call a poison center or physician. Immediately flush eyes with plenty of water, occasionally lifting the upper and lower eyelids. Check for and remove any contact lenses. Continue to rinse for at least 10 minutes. Chemical burns must be treated promptly by a physician.
- Inhalation** : Get medical attention immediately. Call a poison center or physician. Remove victim to fresh air and keep at rest in a position comfortable for breathing. If it is suspected that fumes are still present, the rescuer should wear an appropriate mask or self-contained breathing apparatus. If not breathing, if breathing is irregular or if respiratory arrest occurs, provide artificial respiration or oxygen by trained personnel. It may be dangerous to the person providing aid to give mouth-to-mouth resuscitation. If unconscious, place in recovery position and get medical attention immediately. Maintain an open airway. Loosen tight clothing such as a collar, tie, belt or waistband.

4. First aid measures

- Skin contact** : Get medical attention immediately. Call a poison center or physician. Flush contaminated skin with plenty of water. Remove contaminated clothing and shoes. Wash contaminated clothing thoroughly with water before removing it, or wear gloves. Continue to rinse for at least 10 minutes. Chemical burns must be treated promptly by a physician. Wash clothing before reuse. Clean shoes thoroughly before reuse.
- Ingestion** : Get medical attention immediately. Call a poison center or physician. Wash out mouth with water. Remove dentures if any. Remove victim to fresh air and keep at rest in a position comfortable for breathing. If material has been swallowed and the exposed person is conscious, give small quantities of water to drink. Stop if the exposed person feels sick as vomiting may be dangerous. Do not induce vomiting unless directed to do so by medical personnel. If vomiting occurs, the head should be kept low so that vomit does not enter the lungs. Chemical burns must be treated promptly by a physician. Never give anything by mouth to an unconscious person. If unconscious, place in recovery position and get medical attention immediately. Maintain an open airway. Loosen tight clothing such as a collar, tie, belt or waistband.

Most important symptoms/effects, acute and delayed

Potential acute health effects

- Eye contact** : Causes serious eye damage.
- Inhalation** : May give off gas, vapor or dust that is very irritating or corrosive to the respiratory system.
- Skin contact** : Causes severe burns.
- Ingestion** : May cause burns to mouth, throat and stomach.

Over-exposure signs/symptoms

- Eye contact** : Adverse symptoms may include the following:
pain
watering
redness
- Inhalation** : No specific data.
- Skin contact** : Adverse symptoms may include the following:
pain or irritation
redness
blistering may occur
- Ingestion** : Adverse symptoms may include the following:
stomach pains

Indication of immediate medical attention and special treatment needed, if necessary

- Notes to physician** : Treat symptomatically. Contact poison treatment specialist immediately if large quantities have been ingested or inhaled.
- Specific treatments** : No specific treatment.
- Protection of first-aiders** : No action shall be taken involving any personal risk or without suitable training. If it is suspected that fumes are still present, the rescuer should wear an appropriate mask or self-contained breathing apparatus. It may be dangerous to the person providing aid to give mouth-to-mouth resuscitation. Wash contaminated clothing thoroughly with water before removing it, or wear gloves.

See toxicological information (Section 11)

5. Fire-fighting measures

Extinguishing media

Suitable extinguishing media : Use dry chemical, CO₂, water spray (fog) or foam.

Unsuitable extinguishing media : Do not use water jet.

Specific hazards arising from the chemical : Combustible liquid. In a fire or if heated, a pressure increase will occur and the container may burst, with the risk of a subsequent explosion. Runoff to sewer may create fire or explosion hazard.

Hazardous thermal decomposition products : Decomposition products may include the following materials:
carbon dioxide
carbon monoxide
halogenated compounds
metal oxide/oxides

Special protective actions for fire-fighters : Promptly isolate the scene by removing all persons from the vicinity of the incident if there is a fire. No action shall be taken involving any personal risk or without suitable training. Move containers from fire area if this can be done without risk. Use water spray to keep fire-exposed containers cool.

Special protective equipment for fire-fighters : Fire-fighters should wear appropriate protective equipment and self-contained breathing apparatus (SCBA) with a full face-piece operated in positive pressure mode.

6. Accidental release measures

Personal precautions, protective equipment and emergency procedures

For non-emergency personnel : No action shall be taken involving any personal risk or without suitable training. Evacuate surrounding areas. Keep unnecessary and unprotected personnel from entering. Do not touch or walk through spilled material. Shut off all ignition sources. No flares, smoking or flames in hazard area. Do not breathe vapor or mist. Provide adequate ventilation. Wear appropriate respirator when ventilation is inadequate. Put on appropriate personal protective equipment.

For emergency responders : If specialised clothing is required to deal with the spillage, take note of any information in Section 8 on suitable and unsuitable materials. See also the information in "For non-emergency personnel".

Environmental precautions : Avoid dispersal of spilled material and runoff and contact with soil, waterways, drains and sewers. Inform the relevant authorities if the product has caused environmental pollution (sewers, waterways, soil or air).

Methods and materials for containment and cleaning up

Small spill : Stop leak if without risk. Move containers from spill area. Use spark-proof tools and explosion-proof equipment. Dilute with water and mop up if water-soluble. Alternatively, or if water-insoluble, absorb with an inert dry material and place in an appropriate waste disposal container. Dispose of via a licensed waste disposal contractor.

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6. Accidental release measures

- Large spill** : Stop leak if without risk. Move containers from spill area. Use spark-proof tools and explosion-proof equipment. Approach release from upwind. Prevent entry into sewers, water courses, basements or confined areas. Wash spillages into an effluent treatment plant or proceed as follows. Contain and collect spillage with non-combustible, absorbent material e.g. sand, earth, vermiculite or diatomaceous earth and place in container for disposal according to local regulations (see Section 13). Dispose of via a licensed waste disposal contractor. Contaminated absorbent material may pose the same hazard as the spilled product. Note: see Section 1 for emergency contact information and Section 13 for waste disposal.

7. Handling and storage

Precautions for safe handling

- Protective measures** : Put on appropriate personal protective equipment (see Section 8). Do not get in eyes or on skin or clothing. Do not breathe vapor or mist. Do not ingest. Use only with adequate ventilation. Wear appropriate respirator when ventilation is inadequate. Do not enter storage areas and confined spaces unless adequately ventilated. Keep in the original container or an approved alternative made from a compatible material, kept tightly closed when not in use. Store and use away from heat, sparks, open flame or any other ignition source. Use explosion-proof electrical (ventilating, lighting and material handling) equipment. Use only non-sparking tools. Empty containers retain product residue and can be hazardous. Do not reuse container.

- Conditions for safe storage, including any incompatibilities** : Store in accordance with local regulations. Store in a segregated and approved area. Store in original container protected from direct sunlight in a dry, cool and well-ventilated area, away from incompatible materials (see Section 10) and food and drink. Store locked up. Eliminate all ignition sources. Separate from oxidizing materials. Keep container tightly closed and sealed until ready for use. Containers that have been opened must be carefully resealed and kept upright to prevent leakage. Do not store in unlabeled containers. Use appropriate containment to avoid environmental contamination.

8. Exposure controls/personal protection

Control

Occupational exposure limits

Ingredient name	Exposure limits
potassium hydroxide	ACGIH TLV (United States, 6/2013). C: 2 mg/m ³ OSHA PEL 1989 (United States, 3/1989). CEIL: 2 mg/m ³ NIOSH REL (United States, 10/2013). TWA: 2 mg/m ³ 10 hours.
Ethyl alcohol	ACGIH TLV (United States, 6/2013). STEL: 1000 ppm 15 minutes. OSHA PEL 1989 (United States, 3/1989). TWA: 1000 ppm 8 hours. TWA: 1900 mg/m ³ 8 hours. NIOSH REL (United States, 10/2013). TWA: 1000 ppm 10 hours. TWA: 1900 mg/m ³ 10 hours. OSHA PEL (United States, 2/2013).

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8. Exposure controls/personal protection

Isopropyl alcohol

TWA: 1000 ppm 8 hours.
TWA: 1900 mg/m³ 8 hours.

ACGIH TLV (United States, 6/2013).

TWA: 200 ppm 8 hours.
STEL: 400 ppm 15 minutes.

OSHA PEL 1989 (United States, 3/1989).

TWA: 400 ppm 8 hours.
TWA: 980 mg/m³ 8 hours.
STEL: 500 ppm 15 minutes.
STEL: 1225 mg/m³ 15 minutes.

NIOSH REL (United States, 10/2013).

TWA: 400 ppm 10 hours.
TWA: 980 mg/m³ 10 hours.
STEL: 500 ppm 15 minutes.
STEL: 1225 mg/m³ 15 minutes.

OSHA PEL (United States, 2/2013).

TWA: 400 ppm 8 hours.
TWA: 980 mg/m³ 8 hours.

Appropriate engineering controls

- : Use only with adequate ventilation. Use process enclosures, local exhaust ventilation or other engineering controls to keep worker exposure to airborne contaminants below any recommended or statutory limits. The engineering controls also need to keep gas, vapor or dust concentrations below any lower explosive limits. Use explosion-proof ventilation equipment.

Environmental exposure controls

- : Emissions from ventilation or work process equipment should be checked to ensure they comply with the requirements of environmental protection legislation. In some cases, fume scrubbers, filters or engineering modifications to the process equipment will be necessary to reduce emissions to acceptable levels.

Individual protection measures

Hygiene measures

- : Wash hands, forearms and face thoroughly after handling chemical products, before eating, smoking and using the lavatory and at the end of the working period. Appropriate techniques should be used to remove potentially contaminated clothing. Wash contaminated clothing before reusing. Ensure that eyewash stations and safety showers are close to the workstation location.

Eye/face protection

- : Safety eyewear complying with an approved standard should be used when a risk assessment indicates this is necessary to avoid exposure to liquid splashes, mists, gases or dusts. If contact is possible, the following protection should be worn, unless the assessment indicates a higher degree of protection: chemical splash goggles and/or face shield. If inhalation hazards exist, a full-face respirator may be required instead.

Skin protection

Hand protection

- : Chemical-resistant, impervious gloves complying with an approved standard should be worn at all times when handling chemical products if a risk assessment indicates this is necessary. Considering the parameters specified by the glove manufacturer, check during use that the gloves are still retaining their protective properties. It should be noted that the time to breakthrough for any glove material may be different for different glove manufacturers. In the case of mixtures, consisting of several substances, the protection time of the gloves cannot be accurately estimated.

Body protection

- : Personal protective equipment for the body should be selected based on the task being performed and the risks involved and should be approved by a specialist before handling this product.

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8. Exposure controls/personal protection

- Other skin protection** : Appropriate footwear and any additional skin protection measures should be selected based on the task being performed and the risks involved and should be approved by a specialist before handling this product.
- Respiratory protection** : Use a properly fitted, air-purifying or air-fed respirator complying with an approved standard if a risk assessment indicates this is necessary. Respirator selection must be based on known or anticipated exposure levels, the hazards of the product and the safe working limits of the selected respirator.

9. Physical and chemical properties

Appearance

- Physical state** : Liquid. [Clear.]
- Color** : Red.
- Odor** : soap
- Odor threshold** : Not available.
- pH** : 10.3 to 11.1 [Conc. (% w/w): 100%]
- Melting point** : Not available.
- Boiling point** : Not available.
- Flash point** : Closed cup: 62.8°C (145°F)
- Evaporation rate** : Not available.
- Flammability (solid, gas)** : Not available.
- Lower and upper explosive (flammable) limits** : Not available.
- Vapor pressure** : Not available.
- Vapor density** : Not available.
- Relative density** : 1.024 to 1.034
- Solubility** : Easily soluble in the following materials: cold water and hot water.
- Partition coefficient: n-octanol/water** : Not available.
- Auto-ignition temperature** : Not available.
- Decomposition temperature** : Not available.
- Viscosity** : Not available.

10. Stability and reactivity

- Reactivity** : No specific test data related to reactivity available for this product or its ingredients.
- Chemical stability** : The product is stable.
- Possibility of hazardous reactions** : Under normal conditions of storage and use, hazardous reactions will not occur.
- Conditions to avoid** : Avoid all possible sources of ignition (spark or flame). Do not pressurize, cut, weld, braze, solder, drill, grind or expose containers to heat or sources of ignition.
- Incompatible materials** : Reactive or incompatible with the following materials:
oxidizing materials
- Hazardous decomposition products** : Under normal conditions of storage and use, hazardous decomposition products should not be produced.

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11. Toxicological information

Information on toxicological effects

Acute toxicity

Product/ingredient name	Result	Species	Dose	Exposure
clorofene	LD50 Oral	Rat	1700 mg/kg	-
Ethyl alcohol	LC50 Inhalation Vapor	Rat	124700 mg/m ³	4 hours
	LD50 Oral	Rat	7 g/kg	-
Isopropyl alcohol	LD50 Dermal	Rabbit	12800 mg/kg	-
	LD50 Oral	Rat	5000 mg/kg	-
*Lysol Brand Disinfectant Concentrate, Original Scent	LC50 Inhalation Vapor	Rat	>2.07 mg/l	4 hours
	LD50 Dermal	Rabbit	>5000 mg/kg	-
	LD50 Oral	Rat	5000 mg/kg	-

Conclusion/Summary : Not classified Harmful. *Information is based on toxicity test result of a similar product.

Irritation/Corrosion

Product/ingredient name	Result	Species	Score	Exposure	Observation
clorofene	Skin - Mild irritant	Human	-	48 hours 1 Percent	-
potassium hydroxide	Eyes - Moderate irritant	Rabbit	-	24 hours 1 milligrams	-
	Skin - Severe irritant	Guinea pig	-	24 hours 50 milligrams	-
	Skin - Severe irritant	Human	-	24 hours 50 milligrams	-
	Skin - Severe irritant	Rabbit	-	24 hours 50 milligrams	-
Ethyl alcohol	Eyes - Moderate irritant	Rabbit	-	0.066666667 minutes 100 milligrams	-
	Eyes - Mild irritant	Rabbit	-	24 hours 500 milligrams	-
	Eyes - Moderate irritant	Rabbit	-	100 microliters	-
	Eyes - Severe irritant	Rabbit	-	500 milligrams	-
	Skin - Mild irritant	Rabbit	-	400 milligrams	-
	Skin - Moderate irritant	Rabbit	-	24 hours 20 milligrams	-
Isopropyl alcohol	Eyes - Moderate irritant	Rabbit	-	24 hours 100 milligrams	-
	Eyes - Moderate irritant	Rabbit	-	10 milligrams	-
	Eyes - Severe irritant	Rabbit	-	100 milligrams	-
	Skin - Mild irritant	Rabbit	-	500 milligrams	-
*Lysol Brand Disinfectant Concentrate, Original Scent	Skin - Visible necrosis	Rabbit	-	240 minutes	14 days
	Eyes - Cornea opacity	Rabbit	>3	-	-

Conclusion/Summary

Skin : Causes burns. *Information is based on toxicity test result of a similar product.

Eyes : Causes irreversible eye damage *Information is based on toxicity test result of a similar product.

Sensitization

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11. Toxicological information

Product/ingredient name	Route of exposure	Species	Result
*Lysol Brand Disinfectant Concentrate, Original Scent	skin	Guinea pig	Not sensitizing

Conclusion/Summary

Skin : Non-sensitizer to skin. *Information is based on toxicity test result of a similar product.

Mutagenicity

Not available.

Carcinogenicity

Not available.

Classification

Product/ingredient name	OSHA	IARC	NTP
Ethyl alcohol	-	1	-
Isopropyl alcohol	-	3	-

Reproductive toxicity

Not available.

Teratogenicity

Not available.

Specific target organ toxicity (single exposure)

Name	Category	Route of exposure	Target organs
Isopropyl alcohol	Category 3	Not applicable.	Narcotic effects

Specific target organ toxicity (repeated exposure)

Not available.

Aspiration hazard

Not available.

Information on the likely routes of exposure : Not available.

Potential acute health effects

Eye contact : Causes serious eye damage.

Inhalation : May give off gas, vapor or dust that is very irritating or corrosive to the respiratory system.

Skin contact : Causes severe burns.

Ingestion : May cause burns to mouth, throat and stomach.

Symptoms related to the physical, chemical and toxicological characteristics

Eye contact : Adverse symptoms may include the following:
 pain
 watering
 redness

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11. Toxicological information

- Inhalation** : No specific data.
- Skin contact** : Adverse symptoms may include the following:
pain or irritation
redness
blistering may occur
- Ingestion** : Adverse symptoms may include the following:
stomach pains

Delayed and immediate effects and also chronic effects from short and long term exposure

Short term exposure

- Potential immediate effects** : Not available.
- Potential delayed effects** : Not available.

Long term exposure

- Potential immediate effects** : Not available.
- Potential delayed effects** : Not available.

Potential chronic health effects

Not available.

- General** : No known significant effects or critical hazards.
- Carcinogenicity** : No known significant effects or critical hazards.
- Mutagenicity** : No known significant effects or critical hazards.
- Teratogenicity** : No known significant effects or critical hazards.
- Developmental effects** : No known significant effects or critical hazards.
- Fertility effects** : No known significant effects or critical hazards.

Numerical measures of toxicity

Acute toxicity estimates

Not available.

12. Ecological information

Toxicity

Product/ingredient name	Result	Species	Exposure
clorofene	Acute EC50 0.59 ppm Fresh water	Daphnia - Daphnia magna	48 hours
	Acute LC50 0.33 ppm Fresh water	Fish - Lepomis macrochirus	96 hours
potassium hydroxide	Acute LC50 80 ppm Fresh water	Fish - Gambusia affinis - Adult	96 hours
Ethyl alcohol	Acute EC50 17.921 mg/l Marine water	Algae - Ulva pertusa	96 hours
	Acute EC50 2000 µg/l Fresh water	Daphnia - Daphnia magna	48 hours
	Acute LC50 25500 µg/l Marine water	Crustaceans - Artemia franciscana - Larvae	48 hours
	Acute LC50 42000 µg/l Fresh water	Fish - Oncorhynchus mykiss	4 days
	Chronic NOEC 4.995 mg/l Marine water	Algae - Ulva pertusa	96 hours
	Chronic NOEC 0.375 ul/L Fresh water	Fish - Gambusia holbrooki -	12 weeks

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12. Ecological information

Isopropyl alcohol	Acute LC50 1400000 µg/l Marine water Acute LC50 4200 mg/l Fresh water	Larvae Crustaceans - Crangon crangon Fish - Rasbora heteromorpha	48 hours 96 hours
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Persistence and degradability

Not available.

Bioaccumulative potential

Product/ingredient name	LogP _{ow}	BCF	Potential
clorofene	3.6	-	low
Ethyl alcohol	-0.35	-	low
Isopropyl alcohol	0.05	-	low

Mobility in soil

Soil/water partition coefficient (K_{oc}) : Not available.

Other adverse effects : No known significant effects or critical hazards.

13. Disposal considerations





Disposal methods : The generation of waste should be avoided or minimized wherever possible. Disposal of this product, solutions and any by-products should at all times comply with the requirements of environmental protection and waste disposal legislation and any regional local authority requirements. Dispose of surplus and non-recyclable products via a licensed waste disposal contractor. Waste should not be disposed of untreated to the sewer unless fully compliant with the requirements of all authorities with jurisdiction. Waste packaging should be recycled. Incineration or landfill should only be considered when recycling is not feasible. This material and its container must be disposed of in a safe way. Care should be taken when handling emptied containers that have not been cleaned or rinsed out. Empty containers or liners may retain some product residues. Vapor from product residues may create a highly flammable or explosive atmosphere inside the container. Do not cut, weld or grind used containers unless they have been cleaned thoroughly internally. Avoid dispersal of spilled material and runoff and contact with soil, waterways, drains and sewers.

14. Transport information

Regulatory information	UN number	Proper shipping name	Classes	PG*	Label	Additional information


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14. Transport information

DOT Classification	UN1760	Corrosive liquids, n.o.s. (potassium hydroxide, 2,4-xlenol) RQ (potassium hydroxide, 2,4-xlenol)	8	II		<p>Reportable quantity 25990.9 lbs / 11799.9 kg [3029.3 gal / 11467.3 L] Package sizes shipped in quantities less than the product reportable quantity are not subject to the RQ (reportable quantity) transportation requirements.</p> <p>Limited quantity Yes.</p> <p>Packaging instruction Passenger aircraft Quantity limitation: 1 L</p> <p>Cargo aircraft Quantity limitation: 30 L</p> <p>Special provisions B2, IB2, T11, TP2, TP27</p>
TDG Classification	UN1760	CORROSIVE LIQUID, N.O.S. (potassium hydroxide, 2,4-xlenol)	8	II		<p>Explosive Limit and Limited Quantity Index 1</p> <p>Passenger Carrying Road or Rail Index 1</p> <p>Special provisions 16</p>
Mexico Classification	UN1760	LIQUIDO CORROSIVO, N.E.P. (potassium hydroxide, 2,4-xlenol)	8	II		<p>Special provisions 274</p>
IMDG Class	UN1760	CORROSIVE LIQUID, N.O.S. (potassium hydroxide, 2,4-xlenol)	8	II		<p>Emergency schedules (EmS) F-A, S-B</p> <p>Special provisions 274</p>

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14. Transport information

IATA-DGR Class	UN1760	Corrosive liquid, n.o.s. (potassium hydroxide, 2,4-xyleneol)	8	II		Passenger and Cargo Aircraft Quantity limitation: 1 L Packaging instructions: 851 Cargo Aircraft Only Quantity limitation: 30 L Packaging instructions: 855 Limited Quantities - Passenger Aircraft Quantity limitation: 0.5 L Packaging instructions: Y840 Special provisions A3, A803
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PG* : Packing group

15. Regulatory information

U.S. Federal regulations : TSCA 8(a) PAIR: 2-methylpropan-2-ol
 TSCA 8(a) CDR Exempt/Partial exemption: Not determined
 United States inventory (TSCA 8b): Not determined.
 Clean Water Act (CWA) 307: clorofene; 2,4-xyleneol
 Clean Water Act (CWA) 311: potassium hydroxide; sodium hydroxide; ammonia, anhydrous; xyleneol; m-cresol; p-cresol

Clean Air Act Section 112 (b) Hazardous Air Pollutants (HAPs) : Not listed

Clean Air Act Section 602 Class I Substances : Not listed

Clean Air Act Section 602 Class II Substances : Not listed

DEA List I Chemicals (Precursor Chemicals) : Not listed

DEA List II Chemicals (Essential Chemicals) : Not listed

SARA 302/304

Composition/information on ingredients

Name	%	EHS	SARA 302 TPQ		SARA 304 RQ	
			(lbs)	(gallons)	(lbs)	(gallons)
Ammonia	< 0.01	Yes.	500	-	100	-

SARA 304 RQ : 100000000 lbs / 45400000 kg [11655404.4 gal / 44120505.3 L]

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15. Regulatory information

SARA 311/312

Classification : Fire hazard
Immediate (acute) health hazard

Composition/information on ingredients

Name	%	Fire hazard	Sudden release of pressure	Reactive	Immediate (acute) health hazard	Delayed (chronic) health hazard
clorofene	5 - 10	No.	No.	No.	Yes.	No.
potassium hydroxide	2.5 - 5	No.	No.	No.	Yes.	No.
Ethyl alcohol	1 - 2.5	Yes.	No.	No.	Yes.	No.
Isopropyl alcohol	1 - 2.5	Yes.	No.	No.	Yes.	No.

SARA 313

	Product name	CAS number	%
Form R - Reporting requirements	clorofene	120-32-1	5.4998
	Isopropyl alcohol	67-63-0	1.8333
Supplier notification	clorofene	120-32-1	5.4998
	Isopropyl alcohol	67-63-0	1.8333

SARA 313 notifications must not be detached from the SDS and any copying and redistribution of the SDS shall include copying and redistribution of the notice attached to copies of the SDS subsequently redistributed.

State regulations

Massachusetts : The following components are listed: POTASSIUM HYDROXIDE; ISOPROPYL ALCOHOL; ETHYL ALCOHOL

New York : The following components are listed: Potassium hydroxide

New Jersey : The following components are listed: POTASSIUM HYDROXIDE; CAUSTIC POTASH; ISOPROPYL ALCOHOL; 2-PROPANOL; ETHYL ALCOHOL; ALCOHOL

Pennsylvania : The following components are listed: POTASSIUM HYDROXIDE (K(OH)); CHLORINATED PHENOLS; 2-PROPANOL; DENATURED ALCOHOL

Label elements

Signal word: : DANGER

Hazard statements : Harmful if swallowed.

Corrosive Causes irreversible eye damage
Corrosive CAUSES SKIN BURNS.

Precautionary measures : Keep out of reach of children.
Do not get in eyes, on skin, or on clothing.
Avoid breathing vapor or mist.
Wear protective gloves/protective clothing/eye protection/face protection.
Wash thoroughly with soap and water after handling and before eating, drinking, chewing gum, using tobacco or using the toilet.
Remove contaminated clothing and wash it before reuse.
Avoid breathing dust/fume/gas/mist/vapors/spray.

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16. Other information

Hazardous Material :
Information System (U.S.A.)

Health	3
Flammability	2
Physical hazards	0
Personal protection	D

Caution: HMIS® ratings are based on a 0-4 rating scale, with 0 representing minimal hazards or risks, and 4 representing significant hazards or risks. Although HMIS® ratings are not required on MSDSs under 29 CFR 1910.1200, the preparer may choose to provide them. HMIS® ratings are to be used with a fully implemented HMIS® program. HMIS® is a registered mark of the National Paint & Coatings Association (NPCA). HMIS® materials may be purchased exclusively from J. J. Keller (800) 327-6868.

The customer is responsible for determining the PPE code for this material.

National Fire Protection :
Association (U.S.A.)



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Copyright ©2001, National Fire Protection Association, Quincy, MA 02269. This warning system is intended to be interpreted and applied only by properly trained individuals to identify fire, health and reactivity hazards of chemicals. The user is referred to certain limited number of chemicals with recommended classifications in NFPA 49 and NFPA 325, which would be used as a guideline only. Whether the chemicals are classified by NFPA or not, anyone using the 704 systems to classify chemicals does so at their own risk.

Key to abbreviations :

- ATE = Acute Toxicity Estimate
- BCF = Bioconcentration Factor
- GHS = Globally Harmonized System of Classification and Labelling of Chemicals
- IATA = International Air Transport Association
- IBC = Intermediate Bulk Container
- IMDG = International Maritime Dangerous Goods
- LogPow = logarithm of the octanol/water partition coefficient
- MARPOL 73/78 = International Convention for the Prevention of Pollution From Ships, 1973 as modified by the Protocol of 1978. ("Marpol" = marine pollution)
- UN = United Nations

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Version : 3

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16. Other information

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Revision comments : Update as per US GHS.

✔ Indicates information that has changed from previously issued version.

Notice to reader

To the best of our knowledge, the information contained herein is accurate. However, neither the above-named supplier, nor any of its subsidiaries, assumes any liability whatsoever for the accuracy or completeness of the information contained herein.

Final determination of suitability of any material is the sole responsibility of the user. All materials may present unknown hazards and should be used with caution. Although certain hazards are described herein, we cannot guarantee that these are the only hazards that exist.



RB is a member of the CSPA Product Care Product Stewardship Program.

Safety In The Workplace

Know Your Rights

Under federal law, you are entitled to a safe workplace. Your employer must provide a workplace free of known health and safety hazards. If you have concerns, you have the right to speak up about them without fear of retaliation. You also have the right to:

- Be trained in a language you understand
- Work on machines that are safe
- Be provided required safety gear, such as gloves or a harness and lifeline for falls
- Be protected from toxic chemicals
- Request an OSHA inspection, and speak to the inspector
- Report an injury or illness, and get copies of your medical records
- See copies of the workplace injury and illness log
- Review records of work-related injuries and illnesses
- Get copies of test results done to find hazards in the workplace

When to File a Complaint

- **Safety and Health Complaint**
If you believe working conditions are unsafe or unhealthful, you may file a confidential complaint with OSHA and ask for an inspection. If possible, bring the conditions to your employer's attention.

How to File a Safety and Health Complaint

The Occupational Safety and Health Act of 1970 gives employees and their representatives the right to file a complaint and request an OSHA inspection of their workplace if they believe there is a serious hazard or their employer is not following OSHA standards. Workers do not have to know whether a specific OSHA standard has been violated in order to file a complaint. The complaint should be filed as soon as possible after noticing the hazard or lack of compliance because OSHA citations may only be issued for violations that currently exist or existed in the past 6 months.

Complaints from workers or their representatives are taken seriously by OSHA. OSHA will keep your information confidential. Complaint Filing Options are: Online – Fax/Mail – or Telephone – your local OSHA Regional or Area Office.

Protection from Retaliation

It is illegal for an employer to fire, demote, transfer or otherwise retaliate against a worker for using their rights under the law. If you believe you have been retaliated against in any way, file a whistleblower complaint within 30 days of the alleged retaliation.

What should I do if there is a dangerous situation at work?

If you believe working conditions are unsafe or unhealthful, you may file a complaint with OSHA concerning a hazardous working condition at any time. If possible, bring the conditions to your employer's attention.

What if I am injured on the job?

If you are injured, call a supervisor for help. If the supervisor is not available, get medical assistance or call 911. All employers must notify OSHA within 8 hours of a workplace fatality or within 24 hours of any work-related inpatient hospitalization, amputation or loss of an eye. [Employers under federal OSHA's jurisdiction were required to begin reporting by Jan. 1, 2015. Establishments in a state with a state-run OSHA program may have a different implementation date].

Can someone file a complaint on my behalf?

Yes, a complaint can be filed on your behalf by: an authorized representative of a labor organization or other employee bargaining unit; an attorney; any person acting as a bona fide representative, including members of the clergy, social workers, spouses and other family members; government officials or nonprofit groups; and organizations acting upon specific complaints and injuries from you or your coworkers. In addition, anyone who knows about a workplace safety or health hazard may report unsafe conditions to OSHA, and OSHA will investigate the concerns reported.

What happens after I file a complaint?

Each complaint is evaluated by OSHA to determine whether it should be handled as an off-site investigation or an on-site inspection. Written complaints (or filed online) that are signed by workers or their representative and submitted to an OSHA area or regional office are more likely to result in on-site OSHA inspections.

What are my Employer's responsibilities?

Employer Responsibilities

Under the OSH law, employers have a responsibility to provide a safe workplace. This is a short summary of key employer responsibilities:

- Provide a workplace free from serious recognized hazards and comply with standards, rules and regulations issued under the OSH Act.
- Examine workplace conditions to make sure they conform to applicable OSHA standards.
- Make sure employees have and use safe tools and equipment and properly maintain this equipment.
- Use color codes, posters, labels or signs to warn employees of potential hazards.
- Establish or update operating procedures and communicate them so that employees follow safety and health requirements.
- Employers must provide safety training in a language and vocabulary workers can understand.
- Employers with hazardous chemicals in the workplace must develop and implement a written hazard communication program and train employees on the hazards they are exposed to and proper precautions (and a copy of safety data sheets must be readily available).
- Provide medical examinations and training when required by OSHA standards.
- Post, at a prominent location within the workplace, the OSHA poster (or the state-plan equivalent) informing employees of their rights and responsibilities.
- Report to the nearest OSHA office all work-related fatalities within 8 hours, and all work-related inpatient hospitalizations, all amputations and all losses of an eye within 24 hours.
- Keep records of work-related injuries and illnesses. (Note: Employers with 10 or fewer employees and employers in certain low-hazard industries are exempt from this requirement.

- Provide employees, former employees and their representatives access to the Log of Work-Related Injuries and Illnesses (OSHA Form 300). On February 1, and for three months, covered employers must post the summary of the OSHA log of injuries and illnesses (OSHA Form 300A).
- Provide access to employee medical records and exposure records to employees or their authorized representatives.
- Provide to the OSHA compliance officer the names of authorized employee representatives who may be asked to accompany the compliance officer during an inspection.
- Not discriminate against employees who exercise their rights under the Act. See our "Whistleblower Protection" webpage.
- Post OSHA citations at or near the work area involved. Each citation must remain posted until the violation has been corrected, or for three working days, whichever is longer. Post abatement verification documents or tags.
- Correct cited violations by the deadline set in the OSHA citation and submit required abatement verification documentation.
- OSHA encourages all employers to adopt an Injury and Illness Prevention Program. Injury and Illness Prevention Programs, known by a variety of names, are universal interventions that can substantially reduce the number and severity of workplace injuries and alleviate the associated financial burdens on U.S. workplaces.

What are my rights during an inspection?

When the OSHA inspector arrives, workers and their representatives have the right to talk privately with the OSHA inspector before and after the inspection. A worker representative may also go along on the inspection. Where there is no union or employee representative, the OSHA inspector must talk confidentially with a reasonable number of workers during the course of the investigation.

Does my employer have to provide Personal Protective Equipment (PPE) and who pays for it?

Many OSHA standards require employers to provide personal protective equipment, when it is necessary to protect employees from job-related injuries, illnesses, and fatalities. With few exceptions, OSHA requires employers to pay for personal protective equipment when it is used to comply with OSHA standards. These typically include: hard hats, gloves, goggles, safety glasses, welding helmets and goggles, face shields, chemical protective equipment and fall protection equipment.

Employer's Best Practices

OSHA has recently updated the Guidelines for Safety and Health Programs it first released 30 years ago, to reflect changes in the economy, workplaces, and evolving safety and health issues. The new Recommended Practices have been well received by a wide variety of stakeholders and are designed to be used in a wide variety of small and medium-sized business settings. The Recommended Practices present a step-by-step approach to implementing a safety and health program, built around six core elements that make up a successful program.

The main goal of safety and health programs is to prevent workplace injuries, illnesses, and deaths, as well as the suffering and financial hardship these events can cause for workers, their families, and employers. The recommended practices use a proactive approach to managing workplace safety and health.

Traditional approaches are often reactive –that is, problems are addressed only after a worker is injured or becomes sick, a new standard or regulation is published, or an outside inspection finds a problem that must be fixed. These recommended practices recognize that finding and fixing hazards before they cause injury or illness is a far more effective approach. The idea is to begin with a basic program and simple goals and grow from there. If you focus on achieving goals, monitoring performance, and evaluating outcomes, your workplace can progress along the path to higher levels of safety and health achievement. Employers will find that implementing these recommended practices also brings other benefits. Safety and health programs help businesses:

- Prevent workplace injuries and illnesses
- Improve compliance with laws and regulations
- Reduce costs, including significant reductions in workers' compensation premiums
- Engage workers
- Enhance their social responsibility goals
- Increase productivity and enhance overall business operations

Hazard Identification and Assessment

One of the "root causes" of workplace injuries, illnesses, and incidents is the failure to identify or recognize hazards that are present. or that could have been anticipated. A critical element of any effective safety and health program is a proactive, ongoing process to identify and assess such hazards.

To identify and assess hazards, employers and workers:

- Collect and review information about the hazards present or likely to be present in the workplace.
- Conduct initial and periodic workplace inspections of the workplace to identify new or recurring hazards.
- Investigate injuries, illnesses, incidents, and close calls/near misses to determine the underlying hazards, their causes, and safety and health program shortcomings.
- Group similar incidents and identify trends in injuries, illnesses, and hazards reported.
- Consider hazards associated with emergency or nonroutine situations.
- Determine the severity and likelihood of incidents that could result for each hazard identified, and use this information to prioritize corrective actions.

Action item 1: Collect existing information about workplace hazards Information on workplace hazards may already be available to employers and workers, from both internal and external sources.

How to accomplish it

Collect, organize, and review information with workers to determine what types of hazards may be present and which workers may be exposed or potentially exposed. Information available in the workplace may include:

- Equipment and machinery operating manuals.
- Safety Data Sheets (SDS) provided by chemical manufacturers.
- Self-inspection reports and inspection reports from insurance carriers, government agencies, and consultants.
- Records of previous injuries and illnesses, such as OSHA 300 and 301 logs and reports of incident investigations.

- Workers' compensation records and reports.
- Patterns of frequently-occurring injuries and illnesses.
- Exposure monitoring results, industrial hygiene assessments, and medical records (appropriately redacted to ensure patient/worker privacy).
- Existing safety and health programs (lockout/tagout, confined spaces, process safety management, personal protective equipment, etc.).
- Input from workers, including surveys or minutes from safety and health committee meetings.
- Results of job hazard analyses, also known as job safety analyses.

Information about hazards may be available from outside sources, such as:

- OSHA, National Institute for Occupational Safety and Health (NIOSH), and Centers for Disease Control and Prevention (CDC) websites, publications, and alerts.
- Trade associations.
- Labor unions, state and local occupational safety and health committees/coalitions ("COSH groups"), and worker advocacy groups.
- Safety and health consultants.

Action item 2: Inspect the workplace for safety hazards

Hazards can be introduced over time as workstations and processes change, equipment or tools become worn, maintenance is neglected, or housekeeping practices decline. Setting aside time to regularly inspect the workplace for hazards can help identify shortcomings so that they can be addressed before an incident occurs.

How to accomplish it

- Conduct regular inspections of all operations, equipment, work areas and facilities. Have workers participate on the inspection team and talk to them about hazards that they see or report.
- Be sure to document inspections so you can later verify that hazardous conditions are corrected. Take photos or video of problem areas to facilitate later discussion and brainstorming about how to control them, and for use as learning aids.
- Include all areas and activities in these inspections, such as storage and warehousing, facility and equipment maintenance, purchasing and office functions, and the activities of on-site contractors, subcontractors, and temporary employees.
- Regularly inspect both plant vehicles (e.g., forklifts, powered industrial trucks) and transportation vehicles (e.g., cars, trucks).
- Use checklists that highlight things to look for. Typical hazards fall into several major categories, such as those listed below; each workplace will have its own list:
 - General housekeeping
 - Slip, trip, and fall hazards
 - Electrical hazards
 - Equipment operation
 - Equipment maintenance
 - Fire protection
 - Work organization and process flow (including staffing and scheduling)
 - Work practices
 - Workplace violence

- Ergonomic problems
- Lack of emergency procedures
- Before changing operations, workstations, or workflow; making major organizational changes; or introducing new equipment, materials, or processes, seek the input of workers and evaluate the planned changes for potential hazards and related risks.

Note: Many hazards can be identified using common knowledge and available tools. For example, you can easily identify and correct hazards associated with broken stair rails and frayed electrical cords. Workers can be a very useful internal resource, especially if they are trained in how to identify and assess risks.

Action item 3: Identify health hazards

Identifying workers' exposure to health hazards is typically more complex than identifying physical safety hazards. For example, gases and vapors may be invisible, often have no odor, and may not have an immediately noticeable harmful health effect. Health hazards include chemical hazards (solvents, adhesives, paints, toxic dusts, etc.), physical hazards (noise, radiation, heat, etc.), biological hazards (infectious diseases), and ergonomic risk factors (heavy lifting, repetitive motions, vibration).

Reviewing workers' medical records (appropriately redacted to ensure patient/worker privacy) can be useful in identifying health hazards associated with workplace exposures.

How to accomplish it

- Identify chemical hazards –review SDS and product labels to identify chemicals in your workplace that have low exposure limits, are highly volatile, or are used in large quantities or in unventilated spaces. Identify activities that may result in skin exposure to chemicals.
- Identify physical hazards –identify any exposures to excessive noise (areas where you must raise your voice to be heard by others), elevated heat (indoor and outdoor), or sources of radiation (radioactive materials, X-rays, or radiofrequency radiation).
- Identify biological hazards –determine whether workers may be exposed to sources of infectious diseases, molds, toxic or poisonous plants, or animal materials (fur or scat) capable of causing allergic reactions or occupational asthma.
- Identify ergonomic risk factors –examine work activities that require heavy lifting, work above shoulder height, repetitive motions, or tasks with significant vibration.
- Conduct quantitative exposure assessments –when possible, using air sampling or direct reading instruments.
- Review medical records –to identify cases of musculoskeletal injuries, skin irritation or dermatitis, hearing loss, or lung disease that may be related to workplace exposures.

Note: Identifying and assessing health hazards may require specialized knowledge. Small businesses can obtain free and confidential occupational safety and health advice services, including help identifying and assessing workplace hazards, through OSHA's On-site Consultation Program.

Action item 4: Conduct incident investigations

Workplace incidents –including injuries, illnesses, close calls/near misses, and reports of other concerns– provide a clear indication of where hazards exist. By thoroughly investigating incidents and reports, you will identify hazards that are likely to cause future harm.

The purpose of an investigation must always be to identify the root causes (and there is often more than one) of the incident or concern, in order to prevent future occurrences.

How to accomplish it

- Develop a clear plan and procedure for conducting incident investigations, so that an investigation can begin immediately when an incident occurs. The plan should cover items such as:
 - Who will be involved
 - Lines of communication
 - Materials, equipment, and supplies needed
 - Reporting forms and templates
- Train investigative teams on incident investigation techniques, emphasizing objectivity and open-mindedness throughout the investigation process.
- Conduct investigations with a trained team that includes representatives of both management and workers.
- Investigate close calls/near misses.
- Identify and analyze root causes to address underlying program shortcomings that allowed the incidents to happen.
- Communicate the results of the investigation to managers, supervisors, and workers to prevent recurrence.

Effective incident investigations do not stop at identifying a single factor that triggered an incident. They ask the questions "Why?" and "What led to the failure?" For example, if a piece of equipment fails, a good investigation asks: "Why did it fail?" "Was it maintained properly?" "Was it beyond its service life?" and "How could this failure have been prevented?" Similarly, a good incident investigation does not stop when it concludes that a worker made an error. It asks such questions as: "Was the worker provided with appropriate tools and time to do the work?" "Was the worker adequately trained?" and "Was the worker properly supervised?"

Note: OSHA has special reporting requirements for work-related incidents that lead to serious injury or a fatality (29 CFR 1904.39). OSHA must be notified within 8 hours of a work-related fatality, and within 24 hours of an amputation, loss of an eye, or inpatient hospitalization.

Action item 5: Identify hazards associated with emergency and nonroutine situations

Emergencies present hazards that need to be recognized and understood. Nonroutine or infrequent tasks, including maintenance and startup/shutdown activities, also present potential hazards. Plans and procedures need to be developed for responding appropriately and safely to hazards associated with foreseeable emergency scenarios and nonroutine situations.

How to accomplish it

- Identify foreseeable emergency scenarios and nonroutine tasks, taking into account the types of material and equipment in use and the location within the facility. Scenarios such as the following may be foreseeable:
 - Fires and explosions
 - Chemical releases
 - Hazardous material spills
 - Startups after planned or unplanned equipment shutdowns
 - Nonroutine tasks, such as infrequently performed maintenance activities
 - Structural collapse
 - Disease outbreaks
 - Weather emergencies and natural disasters

- Medical emergencies
- Workplace violence

Action item 6: Characterize the nature of identified hazards, identify interim control measures, and prioritize the hazards for control. The next step is to assess and understand the hazards identified and the types of incidents that could result from worker exposure to those hazards. This information can be used to develop interim controls and to prioritize hazards for permanent control.

How to accomplish it

- Evaluate each hazard by considering the severity of potential outcomes, the likelihood that an event or exposure will occur, and the number of workers who might be exposed.
- Use interim control measures to protect workers until more permanent solutions can be implemented.
- Prioritize the hazards so that those presenting the greatest risk are addressed first. Note, however, that employers have an ongoing obligation to control all serious recognized hazards and to protect workers.

Note: "Risk" is the product of hazard and exposure. Thus, risk can be reduced by controlling or eliminating the hazard or by reducing workers' exposure to hazards.

Personal Protective Equipment

What is personal protective equipment?

Personal protective equipment, commonly referred to as "PPE", is equipment worn to minimize exposure to hazards that cause serious workplace injuries and illnesses. These injuries and illnesses may result from contact with chemical, radiological, physical, electrical, mechanical, or other workplace hazards. Personal protective equipment may include items such as gloves, safety glasses and shoes, earplugs or muffs, hard hats, respirators, or coveralls, vests and full body suits.



*Illustration 1:
Gloves*

What can be done to ensure proper use of personal protective equipment?

All personal protective equipment should be safely designed and constructed, and should be maintained in a clean and reliable fashion. It should fit comfortably, encouraging worker use. If the personal protective equipment does not fit properly, it can make the difference between being safely covered or dangerously exposed.



Illustration 2: Goggles

Employers must provide personal protective equipment to their workers and ensure its proper use. Employers are also required to train each worker required to use personal protective equipment to know:

- When it is necessary
- What kind is necessary
- How to properly put it on, adjust, wear and take it off
- The limitations of the equipment
- Proper care, maintenance, useful life, and disposal of the equipment

HCS Pictograms, Signal Words, and Hazards

As of June 1, 2015, the Hazard Communication Standard (HCS) will require pictograms on labels to alert users of the chemical hazards to which they may be exposed. Each pictogram consists of a symbol on a white background framed within a red border and represents a distinct hazard(s). The pictogram on the label is determined by the chemical hazard classification. The “Signal Word” further describes and labels the classification.

Health Hazard



- Carcinogen
- Mutagenicity
- Reproductive Toxicity
- Respiratory Sensitizer
- Target Organ Toxicity
- Aspiration Toxicity

Flame



- Flammables
- Pyrophorics
- Self-Heating
- Emits Flammable Gas
- Self-Reactives
- Organic Peroxides

Exclamation Mark



- Irritant (skin and eye)
- Skin Sensitizer
- Acute Toxicity (harmful)
- Narcotic Effects
- Respiratory Tract Irritant
- Hazardous to Ozone Layer (Non-Mandatory)

Gas Cylinder



- Gases Under Pressure

Corrosion



- Skin Corrosion/Burns
- Eye Damage
- Corrosive to Metals

Exploding Bomb



- Explosives
- Self-Reactives
- Organic Peroxides

Flame Over Circle



- Oxidizers

Environment (Non-Mandatory)



- Aquatic Toxicity

Skull and Crossbones



- Acute Toxicity (fatal or toxic)

Sample Label





Hazard Communication Standard Labels

OSHA has updated the requirements for labeling of hazardous chemicals under its Hazard Communication Standard (HCS). All labels are required to have pictograms, a signal word, hazard and precautionary statements, the product identifier, and supplier identification. A sample revised HCS label, identifying the required label elements, is shown on the right. Supplemental information can also be provided on the label as needed.



For more information:

SAMPLE LABEL		Hazard Pictograms	
CODE _____ Product Name _____	} Product Identifier	 	Signal Word Danger
Company Name _____ Street Address _____ City _____ State _____ Postal Code _____ Country _____ Emergency Phone Number _____			
Keep container tightly closed. Store in a cool, well-ventilated place that is locked. Keep away from heat/sparks/open flame. No smoking. Only use non-sparking tools. Use explosion-proof electrical equipment. Take precautionary measures against static discharge. Ground and bond container and receiving equipment. Do not breathe vapors. Wear protective gloves. Do not eat, drink or smoke when using this product. Wash hands thoroughly after handling. Dispose of in accordance with local, regional, national, international regulations as specified.		Precautionary Statements Highly flammable liquid and vapor. May cause liver and kidney damage.	
In Case of Fire: use dry chemical (BC) or Carbon Dioxide (CO ₂) fire extinguisher to extinguish. First Aid If exposed call Poison Center. If on skin (or hair): Take off immediately any contaminated clothing. Rinse skin with water.		Hazard Statements Supplemental Information Directions for Use _____ _____ _____ Fill weight: _____ Lot Number: _____ Gross weight: _____ Fill Date: _____ Expiration Date: _____	

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Part 2 Summary

In this segment we have thoroughly discussed important facts about the Occupational Safety and Health Administration's Hazard Communication Standard . We have identified all key elements of Safety Data Sheets and how they apply to businesses. We can now describe factors regarding safety in the workplace and how employers observe safe practices.

Course Summary

In this Course we have thoroughly discussed important facts about Nail Art and topics involving Health, Safety, and Wellbeing. It is recommended that personal service workers, such as salon professionals, stay up to date on any changes or improvements made by OSHA, and Federal and State agencies who regulate safety standards.

Resources and References

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